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(54) Title: HETEROCYCLIC COMPOUNDS

$$R^{1}$$
 $Y$ 
 $Z$ 
 $R^{2}$ 
 $R^{3}$ 
 $R^{3}$ 
 $R^{3}$ 

#### (57) Abstract

Heterocyclic compounds of formula (I), wherein  $R^1$  and  $R^2$  are each lower alkoxy,  $R^3$  is substituted amino, etc., Y is CH or N and Z is CH or N, and pharmaceutically acceptable salts thereof which are useful as a medicament.

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#### DESCRIPTION

#### HETEROCYCLIC COMPOUNDS

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#### TECHNICAL FIELD

This invention relates to new heterocyclic compounds and pharmaceutically acceptable salts thereof which are useful as a medicament.

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#### BACKGROUND ART

Some heterocyclic compounds have been known as described, for example, in U.S. Patent 3,979,516, U.S. Patent 4,021,553, U.S. Patent 4,190,725 and U.S. Patent 4,318,911.

#### DISCLOSURE OF INVENTION

This invention relates to new heterocyclic compounds. More particularly, this invention relates to new heterocyclic compounds and pharmaceutically acceptable salts thereof which have pharmacological activities, processes for preparation thereof, a pharmaceutical composition comprising the same and a use of the same.

Accordingly, one object of this invention is to provide the new and useful heterocyclic compounds and pharmaceutically acceptable salts thereof which possess antithrombotic, vasodilating, and anti-inflammatory activities.

Another object of this invention is to provide processes for preparation of the heterocyclic compounds and salts thereof.

A further object of this invention is to provide a pharmaceutical composition comprising said heterocyclic compounds or a pharmaceutically acceptable salt thereof.

35 Still further object of this invention is to provide

a use of said heterocyclic compound or a pharmaceutically acceptable salt thereof as a medicament for prophylactic and therapeutic treatment of thrombosis, hypertension, cardiovascular or cerebrovascular diseases and inflammation, particularly thrombosis, in human being and animals.

The object heterocyclic compounds of the present invention are novel and can be represented by the following general formula (I):

wherein R<sup>1</sup> and R<sup>2</sup> are each lower alkoxy,
R<sup>3</sup> is heterocyclic group selected from the group

consisting of pyridyl, tetrahydropyridyl, piperidyl, piperazinyl and morpholinyl, which may have suitable substituent(s); substituted amino; carboxy(lower)alkenyl; carboxy(lower)alkyl; hydroxy(lower)alkyl; amino(lower)alkyl which may have suitable substituent(s);

a group of the formula:

0 || -C-R<sup>4</sup>

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(in which  $R^4$  is hydrogen, ethoxy, mono(or di)lower alkylamino, di(lower)alkylamino-(lower)alkylamino, heterocyclicamino 5 which may have suitable substituent(s), or heterocyclic group which may have suitable substituent(s)); or a group of the formula : 10 15 (in which  $R^5$  is lower alkyl or ar(lower)alkyl which may have suitable substituent(s), and X<sup>1</sup> is an acid residue), 20 Y is CH or N and zis CH or N, with proviso that when  $R^3$  is pyridyl; piperidyl which may have hydroxy group; piperazinyl which has lower alkyl group 25 or hydroxy(lower)alkyl group; morpholinyl; lower alkenylamino; hydroxy(lower)alkylamino; phenylamino which may have lower alkoxy group or halogen on the benzene ring; halophenyl(lower)alkylamino; phenylsulfonylamino 30 which has nitro group, amino group or halogen on the benzene ring; or amino substituted with two substituents selected from the group consisting of lower alkyl and hydroxy(lower)alkyl, and 35

Y is N, then Z is CH.

The object compound (I) of the present invention can be prepared by the following processes.

### Process (1)

10  $\mathbb{R}^{1} \longrightarrow \mathbb{C}=0$   $\mathbb{R}^{2} \longrightarrow \mathbb{C}=0$ 15

 $\begin{array}{c|c} & & & \\ & & & \\ & & N \\ & & & \\ & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\$ 

## Process (2)

(IV)
or its reactive derivative
at the carboxy group or
a salt thereof

10 
$$(V)$$
 or a salt thereof

## Process (3)

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elimination reaction of the amino-protective group

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## Process (4)

or its reactive derivative at the amino group , or a salt thereof

or a salt thereof

## Process (5)

## 25 Process (6)

or a salt thereof

or a salt thereof

10

### Process (7)

(Ii) or a salt thereof (Ij) or a salt thereof

### Process (8)

 $\mathbb{R}^1$ 

$$\mathbb{R}^{2}$$

$$\mathbb{R}^{3}$$

$$\mathbb{R}^{3}$$

or a salt thereof

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## Process (9)

(VIII) (IX)

or a salt thereof or a salt thereof

20 or a salt thereof

## Process (10)

25
$$R^1$$
 $Y$ 
 $Z$ 
reduction
 $R^2$ 
 $CH_2OH$ 

(In)
or a salt thereof
 $R^3$ 
or a salt thereof

### Process (11)

### Process (12)

15 
$$R^1$$

Y Z

 $COOH$ 

CH2

COOH

(Io)

(X)

or a salt thereof

25
$$\begin{array}{c}
\mathbb{R}^{1} \\
\mathbb{R}^{2}
\end{array}$$

$$\begin{array}{c}
\mathbb{R}^{2} \\
\mathbb{R}^{2}
\end{array}$$

$$\mathbb{R}^{2} \\
\mathbb{R}^{2} \\
\mathbb{R}^{2}$$

### Process (13)

## Process 14

Process (15)

15
$$R^1$$
 $Y_Z$ 
 $+$ 
 $HN$ 
 $R^{10}$ 
 $R^{11}$ 

20
 $(XIII)$ 
 $(XIV)$ 
or a salt thereof
or a salt thereof

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### Process (16)

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$$\begin{array}{c}
R^{1} \\
Y Z \\
N
\end{array}$$

$$\begin{array}{c}
CH_{2}CH_{2}N \\
R^{13}
\end{array}$$
20
or a salt thereof

## Process (17)

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wherein  $R^1$ ,  $R^2$ ,  $R^3$ ,  $R^5$ , Y, Z and  $X^1$  are each as defined

is mono(or di)lower alkylamino,

di(lower)alkylamino-(lower)alkylamino, heterocyclicamino which may have suitable substituent(s), or heterocyclic group containing at least one nitrogen atom which may have suitable substituent(s),

Ra is protected amino(lower)alkyl,

is amino(lower)alkyl,

is acylamino(lower)alkyl,

 ${
m R}_{\rm d}^3$  is carboxy or protected carboxy,  ${
m R}^6$  and  ${
m R}^7$  are each lower alkyl,

and R<sup>9</sup> are each hydrogen, lower alkyl, cyclo(lower)alkyl, ar(lower)alkyl, heterocyclic group which may have suitable substituent(s), or

R<sup>8</sup> and R<sup>9</sup> are linked together with the attached nitrogen atom to form heterocyclic group which may have suitable substituent(s),

 $x^2$  is a leaving group,

R<sup>10</sup> is hydrogen or lower alkyl and

R<sup>11</sup> is lower alkyl or l-amino-l-iminomethyl, or and R<sup>11</sup> are linked together with the attached

> nitrogen atom to form heterocyclic group which may have suitable substituent(s),

 $R^{12}$  and  $R^{13}$  are each lower alkyl,

Reals carboxy(lower)alken, R<sub>f</sub> is carboxy(lower)alky],

=== in Process (6) means a single or double bond,  $R_{\sigma}^{3}$  is heterocyclic group selected from the group consisting of pyridyl, tetrahydropyridyl,

piperidyl, piperazinyl and morpholinyl,

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which may have suitable substituent(s); substituted amino; carboxy(lower)alkenyl; carboxy(lower)alkyl; hydroxy(lower)alkyl; amino(lower)alkyl which may have suitable substituent(s);

a group of the formula :

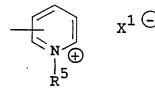
0 || -C-R<sup>4</sup>

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or a group of the formula:



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 $y^1$  is CH or N and  $z^1$  is CH or N.

The starting compound (III) or a salt thereof can be prepared by the following Processes.

### Process (a)

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wherein R<sup>3</sup> is as defined above.

Suitable pharmaceutically acceptable salts of the object compound (I) are conventional non-toxic salts and 20 include a metal salt such as an alkali metal salt (e.g. sodium salt, potassium salt, etc.) and an alkaline earth metal salt (e.g. calcium salt, magnesium salt, etc.), an ammonium salt, an organic base salt (e.g. trimethylamine salt, triethylamine salt, pyridine salt, picoline salt, 25 dicyclohexylamine salt, N,N'-dibenzylethylenediamine salt, etc.), an organic acid salt (e.g. acetate, maleate, tartrate, methanesulfonate, benzenesulfonate, formate, toluenesulfonate, trifluoroacetate, etc.), an inorganic acid salt (e.g. hydrochloride, hydrobromide, sulfate, 30 phosphate, etc.), a salt with an amino acid (e.g. arginine, aspartic acid, glutamic acid, etc.), and the like.

In the above and subsequent descriptions of the present specification, suitable examples and illustrations

of the various definitions which the present invention include within the scope thereof are explained in detail as follows.

The term "lower" is intended to mean 1 to 6 carbon atom(s), preferably one having 1 to 4 carbon atom(s), unless otherwise indicated.

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Suitable "lower alkoxy" may include methoxy, ethoxy, propoxy, isopropoxy, butoxy, isobutoxy, t-butoxy, pentyloxy, t-pentyloxy, hexyloxy and the like.

Suitable "substituent" in the term "heterocyclic group selected from the group consisting of pyridyl, tetrahydropyridyl, piperidyl, piperazinyl and morpholinyl, which may have suitable substituent(s)" may include lower alkyl, ar(lower)alkyl which may have a suitable substituent such as halogen [e.g. fluorine, chlorine, bromine or iodine], and the like.

Suitable "lower alkenyl" in the terms
"carboxy(lower)alkenyl" and "lower alkenylamino" may
include straight or branched one having 2 to 6 carbon
atom(s), such as vinyl, propenyl, butenyl, isobutenyl or
the like, preferably one having 2 to 4 carbon atom(s).

Suitable "substituted amino" may include cyanoamino; imidazolinylamino; guanidino; di(lower)alkylguanidino; lower alkylguanidino; cyclo(lower)alkylguanidino; ar(lower)alkylguanidino; heterocyclicguanidino which may be substituted with suitable substituent(s) such as ar(lower)alkyl or the like;

(1-heterocyclic-l-iminomethyl)amino which may have suitable substituent(s) such as lower alkyl, ar(lower)alkyl, aryl which may have suitable substituent(s) such as lower alkoxy or the like, or the like; di(lower)alkylamino; and the like.

Suitable "lower alkyl" and "lower alkyl moiety" in the terms "carboxy(lower)alkyl", "hydroxy(lower)alkyl",

"amino(lower)alkyl", "di(lower)alkylamino (lower)alkylamino", "acylamino(lower)alkyl", "protected
 amino(lower)alkyl", "di(lower)alkylamino",
 "di(lower)alkylguanidino", "lower alkylguanidino",

5 "ar(lower)alkyl", "hydroxy(lower)alkylamino", "mono(or
 di)lower alkylamino", "halophenyl(lower)alkylamino" and
 "ar(lower)alkylguanidino" may include straight or branched
 one having 1 to 6 carbon atom(s), such as methyl, ethyl,
 propyl, isopropyl, butyl, isobutyl, sec-butyl, t-butyl,

10 pentyl, t-pentyl, hexyl or the like, preferably one having
 1 to 4 carbon atom(s).

Suitable "protected amino moiety" in the term "protected amino(lower)alkyl" may include acylamino and the like.

Suitable "protected carboxy" may include esterified carboxy and the like.

Suitable example of the ester moiety of an esterified carboxy may be the ones such as lower alkyl ester (e.g. methyl ester, ethyl ester, propyl ester, isopropyl ester, butyl ester, isobutyl ester, tert-butyl ester, pentyl ester, hexyl ester, l-cyclopropylethyl ester, etc.) which may have at least one suitable substituent(s), for example, lower alkanoyloxy(lower)alkyl ester [e.g. acetoxymethyl ester, propionyloxymethyl ester, butyryloxymethyl ester, valeryloxymethyl ester, pivaloyloxymethyl ester, hexanoyloxymethyl ester, l(or 2)-acetoxyethyl ester, l(or 2 or 3)-acetoxypropyl ester, l(or 2 or 3 or 4)-acetoxybutyl ester, l(or

2)-propionyloxyethyl ester, l(or 2 or

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3)-propionyloxypropyl ester, 1(or 2)-butyryloxyethyl ester, 1(or 2)-isobutyryloxyethyl ester, 1(or 2)-pivaloyloxyethyl ester, 1(or 2)-hexanoyloxyethyl ester, isobutyryloxymethyl ester, 2-ethylbutyryloxymethyl ester, 3,3-dimethylbutyryloxymethyl ester, 1(or 2)-pentanoy oxyethyl ester, etc.], lower alkanesulfonyl(lower)alkyl ester

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(e.g. 2-mesylethyl ester, etc.), mono(or di or tri)halo(lower)alkyl ester (e.g. 2-iodoethyl ester, 2,2,2-trichloroethyl ester, etc.), lower alkoxycarbonyloxy(lower)alkyl ester (e.g. methoxycarbonyloxymethyl ester, ethoxycarbonyloxymethyl ester, 5 2-methoxycarbonyloxyethyl ester, 1-ethoxycarbonyloxyethyl ester, 1-isopropoxycarbonyloxyethyl ester, etc.), phthalidylidene(lower)alkyl ester, or (5-lower alkyl 2-oxo-1,3-dioxol-4-yl)(lower)alkyl ester [e.g. (5-methyl-2-oxo-1,3-dioxol-4-yl)methyl ester, 10 (5-ethyl-2-oxo-1,3-dioxol-4-yl)methyl ester, (5-propyl-2-oxo-1,3-dioxol-4-yl)ethyl ester, etc.]; lower alkenyl ester (e.g. vinyl ester, allyl ester, etc.); lower alkynyl ester (e.g. ethynyl ester, propynyl ester, 15 etc.): ar(lower)alkyl ester which may have at least one suitable substituent(s) such as mono(or di or tri)-phenyl(lower)alkyl ester which may have at least one suitable substituent(s) (e.g. benzyl ester, 4-methoxybenzyl ester, 4-nitrobenzyl ester, phenethyl ester, trityl ester, 20 benzhydryl ester, bis(methoxyphenyl)methyl ester, 3,4-dimethoxybenzyl ester, 4-hydroxy-3,5-di-tertbutylbenzyl ester, etc.); aryl ester which may have at least one suitable substituent(s) (e.g. phenyl ester, 4-chlorophenyl ester, 25 tolyl ester, tert-butylphenyl ester, xylyl ester, mesityl ester, cumenyl ester, etc.); phthalidyl ester; and the like. Suitable "acyl" in the terms "acylamino" and

"acylamino(lower)alkyl" may include carbamoyl, aliphatic acyl group and acyl group containing an aromatic ring, which is referred to as aromatic acyl, or heterocyclic ring which is referred to as heterocyclic acyl.

Suitable example of said acyl may be illustrated as follows:-

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Carbamoyl;
      Alliphatic acyl such as lower or higher alkanoyl (e.g.
      formyl, acetyl, propanoyl, butanoyl, 2-methylpropanoyl,
      pentanoy1, 2,2-dimethylpropanoy1, hexanoy1, heptanoy1,
      octanoyl, nonanoyl, decanoyl, undecanoyl, dodecanoyl,
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      tridecanoyl, tetradecanoyl, pentadecanoyl, hexadecanoyl,
      heptadecanoyl, octadecanoyl, nonadecanoyl, icosanoyl,
      etc.);
      lower or higher alkoxycarbonyl (e.g. methoxycarbonyl,
      ethoxycarbonyl, t-butoxycarbonyl, t-pentyloxycarbonyl,
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      heptyloxycarbonyl, etc.);
      lower alkylcarbamoyl (e.g. methylcarbamoyl,
      ethylcarbamoyl, propylcarbamoyl, isopropylcarbamoyl,
      butylcarbamoyl, etc.), lower or higher alkanesulfonyl
      (e.g. methanesulfonyl, ethanesulfonyl, etc.);
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      lower or higher alkoxysulfonyl (e.g. methoxysulfonyl,
      ethoxysulfonyl, etc.); or the like;
           Aromatic acyl such as
           aroyl (e.g. benzoyl, toluoyl, naphthoyl, etc.);
           ar(lower)alkanoyl [e.g. phenyl(lower)alkanoyl (e.g.
20
      phenylacetyl, phenylpropanoyl, phenylbutanoyl,
      phenylisobutylyl, phenylpentanoyl, phenylhexanoyl, etc.),
      naphthyl(lower)alkanoyl (e.g. naphthylacetyl,
      naphthylpropanoyl, naphthylbutanoyl, etc.), etc.];
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           ar(lower)alkenoyl [e.g. phenyl(lower)alkenoyl (e.g.
      phenylpropenoyl, phenylbutenoyl, phenylmethacryloyl,
      phenylpentenoyl, phenylhexenoyl, etc.),
      naphthyl(lower)alkenoyl (e.g. naphthylpropenoyl,
      naphthylbutenoyl, naphthylpentenoyl, etc.), etc.];
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           ar(lower)alkoxycarbonyl [e.g. phenyl(lower)alkoxy-
      carbonyl (e.g. benzyloxycarbonyl, etc.), etc.];
      aryloxycarbonyl (e.g. phenoxycarbonyl,
      naphthyloxycarbonyl, etc.);
      aryloxy(lower)alkanoyl (e.g. phenoxyacetyl,
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      phenoxypropionyl, etc.);
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arylcarbamoyl (e.g. phenylcarbamoyl, etc.);
 arylthiocarbamoyl (e.g. phenylthiocarbamoyl, etc.);
 arylglyoxyloyl (e.g. phenylglyoxyloyl, naphthylglyoxyloyl,
 etc.);
 arenesulfonyl (e.g. benzenesulfonyl, p-toluenesulfonyl,
 etc.); or the like;
      Heterocyclic acyl such as
 heterocycliccarbonyl;
 heterocyclic (lower)alkanoyl (e.g. thienylacetyl,
 thienylpropanoyl, thienylbutanoyl, thienylpentanoyl,
thienylhexanoyl, thiazolylacetyl, thiadiazolylacetyl,
tetrazolylacetyl, etc.);
heterocyclic(lower)alkenoyl (e.g. heterocyclicpropenoyl,
heterocyclicbutenoyl, heterocyclicpentenoyl,
heterocyclichexenoyl, etc.);
heterocyclicglyoxyloyl (e.g. thiazolylglyoxyloyl,
thienylglyoxyloyl, etc.); or the like; in which suitable
hetetocyclic moiety in the terms "hetetocycliccarbonyl",
heterocyclic(lower)alkanoyl", heterocyclic(lower)alkenoyl
and "heterocyclicglyoxyloyl" as mentioned above means, in
more detail, saturated or unsaturated, monocyclic or
polycyclic heterocyclic group containing at least one
hetero-atom such as an oxygen, sulfur, nitrogen atom and
the like. And, especially preferable heterocyclic group
may be heterocyclic group such as
     unsaturated 3 to 8-membered more preferably 5 or
6-membered heteromonycyclic group containing 1 to
4-nitrogen atom(s), for example, pyrrolyl, pyrrolinyl,
imidazolyl, pyrazolyl, pyridyl and its N-oxide,
dihydropyridyl, pyrimidyl, pyrazinyl, pyridazinyl,
tetrahydropyridazinyl (e.g. 2,3,4,5-tetrahydropyridazinyl,
etc.), triazolyl (e.g. 4H-1,2,4-triazolyl,
1H-1,2,3-triazolyl, 2H-1,2,3-triazolyl, etc.), tetrazolyl
(e.g. lH-tetrazolyl, 2H-tetrazolyl, etc.), etc.;
```

saturated 3 to 8-membered (more preferably 5 or 6-

membered)heteromonocyclic group containing 1 to 4 nitrogen atom(s), for example pyrrolidinyl, imidazolidinyl, piperidino, piperazinyl, etc.; unsaturated condensed heterocyclic group containing 1 to 4 nitrogen atom(s), for example, indolyl, isoindolyl, 5 indolinyl, indolizinyl, benzimidazolyl, quinolyl, isoquinolyl, indazolyl, benzotriazolyl, etc.; unsaturated 3 to 8-membered (more preferably 5 or 6membered)heteromonocyclic group containing 1 to 2 oxygen atom(s) and 1 to 3 nitrogen atom(s), for example, 10 oxazolyl, isoxazolyl, oxadiazolyl (e.g. 1,2,4-oxadiazolyl, 1,3,4-oxadiazolyl, 1,2,5-oxadiazolyl, etc.) etc.; saturated 3 to 8-membered (more preferably 5 or 6membered)heteromonocyclic group containing 1 to 2 oxygen atom(s) and 1 to 3 nitrogen atom(s), for example, 15 morpholinyl, sydnonyl, etc.; unsaturated condensed heterocyclic group containing 1 to 2 oxygen atom(s) and 1 to 3 nitrogen atom(s), for example, benzoxazolyl, benzoxadiazolyl, etc.; unsaturated 3 to 8-membered (more preferably 5 or 20 6-membered)heteromonocyclic group containing 1 to 2 sulfur atom(s) and 1 to 3 nitrogen atom(s), for example, thiazolyl, isothiazolyl, thiadiazolyl (e.g. 1.2.3-thiadiazolyl, 1,2 4-thiadiazolyl, 1.3.4-thiadiazolyl, 1,2,5-thiadiazolyl, etc.), 25 dihydrothiazinyl, etc.; saturated 3 to 8-membered (more preferably 5 or 6-membered) heteromonocyclic group containing 1 to 2 sulfur atom(s) and 1 to 3 nitrogen atom(s), for example, thiazolidinyl, etc.; 30 unsaturated 3 to 8-membered (more preferably 5 or 6-membered)heteromonocyclic group containing 1 to 2 sulfur atom(s), for example, thienyl, dihydrodithiinyl, dihydrodithionyl, etc.;

unsaturated condensed heterocyclic group containing 1 to 2

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sulfur atom(s) and 1 to 3 nitrogen atom(s), for example, benzothiazoly1, benzothiadiazoly1, etc.; unsaturated 3 to 8-membered (more preferably 5 to 6-membered) heteromonocyclic group containing an oxygen atom, for example, fury1, etc.; unsaturated 3 to 8-membered (more preferably 5 or 6-membered) heteromonocyclic group containing an oxygen atom and 1 to 2 sulfur atom(s), for example, dihydrooxathiiny1, etc.;

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unsaturated condensed heterocyclic group containing 1 to 2 sulfur atom(s), for example, benzothienyl, benzodithiinyl, etc.;

unsaturated condensed heterocyclic group containing an oxygen atom and 1 to 2 sulfur atom(s), for example,

- benzoxathiinyl, etc. and the like. The acyl moiety as stated above may have one to ten, same or different, suitable substituent(s) such as halogen (e.g. fluorine, chlorine, bromine or iodine), hydroxy, nitro, oxo, lower alkyl (e.g. methyl, ethyl propyl, isopropyl, butyl, isobutyl, t-butyl, pentyl, hexyl, etc.) or the like.
- Suitable "heterocyclic group" in the terms

  "heterocyclic group which may have suitable
  substituent(s)", "(1-heterocyclic-1-iminomethyl)amino"

  "heterocyclicguanidino" and "heterocyclicamino which may
  have suitable substituent(s)" can be referred to the ones
  as mentioned above.

Suitable "substituent" in the term "heterocyclic group which may have suitable substituent(s)" may include lower alkyl (e.g. methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, t-butyl, pentyl, hexyl, etc.) ar(lower)alkyl, hydroxy(lower)alkyl and the like.

Suitable "heterocyclic group" in the definitions " $\mathbb{R}^8$  and  $\mathbb{R}^9$  are linked together with the attached nitrogen atom to form heterocyclic group which may have suitable substituent(s)", " $\mathbb{R}^{10}$  and  $\mathbb{R}^{11}$  are linked together with the

attached nitrogen atom to form heterocyclic group which may have suitable substituent(s)" and "heterocyclic group containing at least one nitrogen atom which may have suitable substituent(s)" may include saturated 3 to 8-membered (more preferably 5 or 6-membered) heteromonocyclic group containing 1 to 4 nitrogen atom(s), for example pyrrolidinyl, imidazolidinyl, piperidino, piperazinyl, etc.;

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saturated 3 to 8-membered (more preferably 5 or 6-membered) heteromonocyclic group containing 1 to 2 oxygen atom(s) and 1 to 3 nitrogen atom(s), for example, morpholinyl, sydnonyl, etc. and the like.

Suitable "substituent" in the term "heterocyclicamino which may have suitable substituent(s)" may include ar(lower)alkyl, and the like.

Suitable "substituent" in the definitions "R<sup>8</sup> and R<sup>9</sup> are linked together with the attached nitrogen atom to form heterocyclic group which may have suitable substituent(s)" and "R<sup>10</sup> and R<sup>11</sup> are linked together with the attached nitrogen atom to form heterocyclic group which may have suitable substituent(s)" may include lower alkyl, ar(lower)alkyl, aryl having lower alkoxy, and the like.

Suitable "substituent" in the definition

"heterocyclic group containing at least one nitrogen atom
which may have suitable substituent(s)" may include lower
alkyl, ar(lower)alkyl, hydroxy(lower)alkyl, and the like.

Suitable "cyclo(lower)alkyl" and "cyclo(lower)alkyl moiety" in the term "cyclo(lower)alkylguanidino" may include 3 to 8-membered cycloalkyl such as cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cyclohexyl, cycloheptyl, cyclooctyl, and the like, preferably one having 5 to 7 carbon atoms.

Suitable "substituent" in the term "amino(lower)alkyl which may have suitable substituent(s)" may include lower

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alkyl, acyl as exemplified above, and the like.

Suitable " acid residue" may include halogen as exemplified above.

Suitable "leaving group" may include an acid residue as exemplified above, and the like.

Suitable "aryl" and "aryl moiety" in the terms "ar(lower)alkyl" and "ar(lower)alkylguanidino" may include phenyl, naphthyl and the like.

Suitable "substituent" in the term "ar(lower)alkyl which may have suitable substituent(s)" may include halogen as exemplified above, and the like.

Suitable "halogen" and "halogen moiety" in the term "halophenyl(lower)alkylamino" is as exemplified above.

Preferred embodiments of the object compound (I) are as follows.

R<sup>1</sup> is lower alkoxy, R<sup>2</sup> is lower alkoxy,

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 $\mathbb{R}^3$  is heterocyclic group selected from the group 20 consisting of pyridyl, tetrahydropyridyl, piperidyl, piperazinyl and morpholinyl, which may have one to three (more preferably one or two) suitable substituent(s) [more preferably heterocyclic group 25 selected from the group consisting of pyridyl, tetrahydropyridyl, piperidyl, piperazinyl and morpholinyl, which may have one or two substituent(s) selected from the group consisting of lower alkyl and ar(lower)alkyl which may have one or two halogen; 30 most preferably heterocyclic group selected from the group consisting of pyridyl, tetrahydropyridyl, piperidyl, piperazinyl and morpholinyl, which may have one or two substituent(s) selected from the group consisting of lower alkyl, phenyl(lower)alkyl 35 and halophenyl(lower)alkyl];

cyanoamino; imidazolinylamino; guanidino; di(lower)alkylguanidino; lower alkylguanidino; cyclo(lower)alkylguanidino; ar(lower)alkylguanidino [more preferably phenyl(lower)alkylguanidino]; heterocyclicquanidino which may be substituted with 5 one to three (more preferably one) suitable substituent(s) [more preferably heterocyclicguanidino which may have ar(lower)alkyl; most preferably phenyl(lower)alkylpiperidylguanidino]; 10 (1-heterocyclic-1-iminomethyl)amino which may have one to three (more preferably one) suitable substituent(s) [more preferably (1-heterocyclic-1-iminomethyl)amino which may have substituent selected from the group consisting of lower alkyl, ar(lower)alkyl and aryl which may have 15 lower alkoxy; most preferably {morpholinyl(imino)methyl}amino, {piperidyl(imino)methyl}amino which may have phenyl(lower)alkyl, or 20 {piperazinyl(imino)methyl}amino which may have substituent selected from the group consisting of lower alkyl, phenyl(lower)alkyl and lower alkoxy phenyl]; di(lower)alkylamino; hydroxy(lower)alkyl; carboxy(lower)alkyl; carboxy(lower)alkenyl; 25 amino(lower)alkyl which may have one to three (more preferably one or two) substituent(s) selected from the group consisting of lower alkyl and acyl [more preferably amino(lower)alkyl which may have one or two substituent(s) selected from the group consisting 30 of lower alkyl, lower alkanoyl, lower alkylcarbamoyl, lower alkoxycarbonyl and heterocycliccarbonyl which may have one to three suitable substituent(s); most preferably amino(lower)alkyl which may have one or two substituent(s) selected from the group consisting 35 of lower alkyl, lower alkanoyl, lower alkylcarbamoyl,

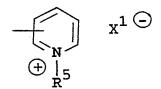
lower alkoxycarbonyl and tetrahydropyridazinylcarbonyl which may have oxo group];
a group of the formula :

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(in which R4 is hydrogen, ethoxy, mono(or di)lower alkylamino, 10 di(lower)alkylamino-(lower)alkylamino, heterocyclicamino which may have one to three (more preferably one or two) substituent(s) selected from the group consisting of lower alkyl, lower 15 alkoxy, halogen and ar(lower)alkyl [more preferably saturated 5 or 6-membered heteromonocyclicamino in which heteromonocyclic group contains 1 to 3 nitrogen atom(s), which may 20 have one or two substituent(s) selected from the group consisting of lower alkyl, lower alkoxy halogen and phenyl(lower)alkyl; most preferably phenyl(lower)alkylpiperidylamino] or 25 heterocyclic group which may have one to three (more preferably one or two) substituent(s) selected from the group consisting of lower alkyl, lower alkoxy, halogen, ar(lower)alkyl and 30 hydroxy(lower)alkyl [more preferably saturated 5 or 6-membered heteromonocyclic group containing 1 to 2 oxygen atom(s) and 1 to 3 nitrogen atom(s), which may have one or two 35 substituent(s) selected from the group

consisting of lower alkyl, lower
alkoxy, halogen, phenyl(lower)alkyl
and hydroxy(lower)alkyl, or saturated
5 or 6-membered heteromonocyclic group
containing 1 to 3 nitrogen atom(s),
which may have one or two
substituent(s) selected from the group
consisting of lower alkyl, lower
alkoxy, halogen, phenyl(lower)alkyl
and hydroxy(lower)alkyl; most
preferably morpholinyl, or piperazinyl
which may have lower alkyl,
hydroxy(lower)alkyl or
phenyl(lower)alkyl);

or a group of the formula:



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(in which R<sup>5</sup> is lower alkyl, or ar(lower)alkyl which may have one to three (more preferably one or two)substituent(s) selected from the group consisting of 25 halogen, lower alkyl and lower alkoxy [more preferably phenyl(lower)alkyl which may have one or two substituent(s) selected from the group consisting of halogen, lower alkyl and 30 lower alkoxy; most preferably phenyl(lower)alkyl which may have halogen], and x1 is an acid residue [more preferably halogen]),

35 Y is CH or N and

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Z is CH or N, with proviso that

when R<sup>3</sup> is pyridyl; piperidyl which may have hydroxy
group; piperazinyl which has lower alkyl group
or hydroxy(lower)alkyl group; morpholinyl; lower
alkenylamino; hydroxy(lower)alkylamino;
phenylamino which may have lower alkoxy group or
halogen on the benzene ring;
halophenyl(lower)alkylamino; phenylsulfonylamino
which has nitro group, amino group or halogen on
the benzene ring; or amino substituted with two

Y is N,

15 then Z is CH.

The processes for preparing the object and starting compounds are explained in detail in the following.

substituents selected from the group consisting

of lower alkyl and hydroxy(lower)alkyl, and

### 20 Process (1)

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The compound (Ia) or a salt thereof can be prepared by reacting the compound (II) with the compound (III) or a salt thereof.

The reaction is usually carried out in a conventional solvent such as water, alcohols (e.g., methanol, ethanol, isopropyl alcohol, etc.), tetrahydrofuran, dioxane, chloroform, methylene chloride, dimethyl acetamide, N,N-dimethylformamide or any other organic solvent which does not adversely influence the reaction. Among these solvents, hydrophilic solvents may be used in a mixture with water.

This reaction is preferably carried out in the presence of an inorganic or an organic acid such as hydrochloric acid, hydrobromic acid, sulfuric acid, etc.), an organic acid (e.g. formic acid, acetic acid,

trifluoroacetic acid, propionic acid, methanesulfonic acid, benzenesulfonic acid, p-toluenesulfonic acid, etc.), and the like.

The reaction may be also carried out in the presence of an inorganic or an organic base such as an alkali metal hydroxide, an alkali metal hydrogencarbonate, alkali metal carbonate, alkali metal acetate, tri(lower)alkylamine, pyridine (e.g. pyridine, lutidine, picoline, dimethylaminopyridine, etc.), N-(lower)alkylmorpholine, N,N-di(lower)alkylbenzylamine, N,N-di(lower)alkylaniline or the like. When the base, the acid and/or the starting compound are in liquid, they can be used also as a solvent.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

Suitable salts of the compounds (III) and (Ia) can be referred to the ones as exemplified for the compound (I).

#### Process (2)

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20 The compound (Ib) or a salt thereof can be prepared by reacting a compound (IV) or its reactive derivative at the carboxy group or a salt thereof with a compound (V) or a salt thereof.

As suitable said reactive derivatives at the carboxy group, there may be mentioned acid halides, acid anhydrides, active amides and esters. Suitable examples are acid halides such as acid chloride and acid bromide, mixed acid anhydrides with various acids [e.g. substituted phosphoric acid such as dialkyl phosphoric acid, sulfuric acid, aliphatic carboxylic acid, aromatic carboxylic acid, etc.], symmetric acid anhydrides, active amides with various imidazoles, and esters such as lower alkyl ester [e.g. methyl ester, ethyl ester, etc.], cyanomethyl ester, methoxymethyl ester, p-nitrophenyl ester,

35 2,4-dinitrophenyl ester, pentachlorophenyl ester,

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phenylazophenyl ester, carboxymethylthio ester, and N-hydroxysuccinimide ester.

The reaction is usually carried out in a conventional solvent, such as methylene chloride, chloroform, alcohol [e.g. methanol, ethanol, etc.], benzene, toluene, pyridine, diethyl ether, dioxane, tetrahydrofuran, acetone, acetonitrile, ethyl acetate,

N,N-dimethylformamide or any other organic solvent which does not adversely affect the reaction. In case that the compound (V) is liquid, it can also be used as a solvent.

The reaction temperature is not critical and the reaction can be carried out under cooling, at ambient temperature, or under heating.

This reaction can typically be conducted in the presence or absence of an accelerator such as base.

Suitable base may include a tertiary amine [e.g. triethylamine, pyridine, N,N-dimethylaniline, etc.], an alkali metal hydroxide [e.g. sodium hydroxide, potassium hydroxide, etc.], an alkali metal carbonate [e.g. sodium carbonate, potassium carbonate, etc.], alkali metal bicarbonate [e.g. sodium bicarbonate, etc.], a salt of an organic acid [e.g. sodium acetate, etc.] and the like. Ir case that the base is liquid, the base can be used as a solvent.

Suitable salts of the compounds (IV), (V) and (Ib) can be referred to the ones as exemplified for the compound (I).

### Process (3)

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The compound (Id) or a salt thereof can be prepared by subjecting the compound (Ic) or a salt thereof to elimination reaction of the amino-protective group on R<sub>a</sub>. Suitable method of this reaction may include conventional one such as hydrolysis, reduction and the like.

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### (i) For Hydrolysis:

The hydrolysis is preferably carried out in the presence of a base or an acid including Lewis acid.

Suitable base may include an inorganic base and an organic base such as an alkali metal [e.g. sodium, potassium, etc.], the hydroxide or carbonate or bicarbonate thereof, trialkylamine [e.g. trimethylamine, triethylamine, etc.], picoline, 1,5-diazabicyclo[4.3.0]-non-5-ene, 1,4-diazabicyclo[2.2.2]octane,

1,8-diazabicyclo[5.4.0]undec-7-ene, or the like.

Suitable acid may include an organic acid [e.g. formic acid, acetic acid, propionic acid, trichloroacetic acid, trifluoroacetic acid, etc.] and an inorganic acid [e.g. hydrochloric acid, hydrobromic acid, sulfuric acid, hydrogen chloride, hydrogen bromide, etc.]. The elimination using Lewis acid such as trihaloacetic acid [e.g. trichloroacetic acid, trifluoroacetic acid, etc.] or the like is preferably carried out in the presence of cation trapping agents [e.g. anisole, phenol, etc.].

The reaction is usually carried out in a solvent such as water, an alcohol [e.g. methanol, ethanol, etc.], methylene chloride, tetrahydrofuran, a mixture thereof or any other solvent which does not adversely influence the reaction. A liquid base or acid can be also used as the solvent. The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

### (ii) For reduction:

Reduction is carried out in a conventional manner, including chemical reduction and catalytic reduction.

Suitable reducing agents to be used in chemical reduction are a combination of a metal (e.g. tin, zinc, iron, etc.) or metallic compound (e.g. chromium chloride, chromium acetate, etc.) and an organic or inorganic acid (e.g. formic acid, acetic acid, propionic acid,

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trifluoroacetic acid, p-toluenesulfonic acid, hydrochloric acid, hydrobromic acid, etc.).

Suitable catalysts to be used in catalytic reduction are conventional ones such as platinum catalysts (e.g. 5 platinum plate, spongy platinum, platinum black, colloidal platinum, platinum oxide, platinum wire, etc.), palladium catalysts (e.g. spongy palladium, palladium black, palladium oxide, palladium on carbon, colloidal palladium palladium on barium sulfate, palladium on barium 10 carbonate, etc.), nickel catalysts (e.g. reduced nickel, nickel oxide, Raney nickel, etc.), cobalt catalysts (e.g. reduced cobalt, Raney cobalt, etc.), iron catalysts (e.g. reduced iron, Raney iron, etc.), copper catalysts (e.g. reduced copper, Raney copper, Ullman copper, etc.) and the 15 The reduction is usually carried out in a conventional solvent which does not adversely influence the reaction such as water, methanol, ethanol, propanol, N,N-dimethylformamide, tetrahydrofuran, or a mixture thereof. Additionally, in case that the above-mentioned 20 acids to be used in chemical reduction are in liquid, they can also be used as a solvent.

The reaction temperature of this reduction is not critical and the reaction is usually carried out under cooling to warming.

Suitable salts of the compounds (Ic) and (Id) can be referred to the ones as exemplified for the compound (I).

#### Process (4)

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The compound (Ie) or a salt thereof can be prepared by subjecting the compound (Id) or its reactive derivative at the amino group or a salt thereof to acylation reaction.

Suitable acylating agent to be used in the present acylation reaction may include the compound of the formula:

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 $R^{14}$  - OH (XXII)

(wherein R<sup>14</sup> is acyl)

or its reactive derivative or a salt thereof.

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Suitable reactive derivative at the amino group of the compound (Id) may include Schiff's base type imino or its tautomeric enamine type isomer formed by the reaction of the compound (Id) with a carbonyl compound such as aldehyde, ketone or the like; a silyl derivative formed by the reaction of the compound (Id) with a silyl compound such as N,O-bis(trimethylsilyl)acetamide,
N-trimethylsilylacetamide or the like; a derivative formed by the reaction of the compound (Id) with phosphorus trichloride or phosgene, and the like.

Suitable salts of the compound (Id) and (Ie) can be referred to the ones as exemplified for the compound (I).

Suitable reactive derivative of the compound (XXII) may include an acid halide, an acid anhydride, an 20 activated amide, an activated ester, isocyanate, and the like. The suitable example may be an acid chloride an acid azide; a mixed acid anhydride with an acid such as substituted phosphoric acid (e.g. dialkylphosphoric acid, phenylphosphoric acid, diphenylphosphoric acid, 25 dibenzylphosphoric acid, halogenated phosphoric acid, etc.), dialkylphosphorous acid, sulfurous acid, thiosulfuric acid, alkanesulfonic acid (e.g. methanesulfonic acid, ethanesulfonic acid, etc.), sulfuric acid, alkylcarbonic acid, aliphatic carboxylic acid (e.g. 30 pivalic acid, pentanoic acid, isopentanoic acid, 2-ethylbutyric acid or trichloroacetic acid, etc.) or aromatic carboxylic acid (e.g. benzoic acid, etc.); a symmetrical acid anhydride; an activated amide with imidazole, 4-substituted imidazole, dimethylpyrazole, 35

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triazole or tetrazole; or an activated ester (e.g. cyanomethyl ester, methoxymethyl ester, dimethyliminomethyl [(CH $_3$ ) $_2$  $\vec{N}$ =CH-] ester, vinyl ester, propargyl ester, p-nitrophenyl ester, 2,4-dinitrophenyl ester, trichlorophenyl ester, pentachlorophenyl ester, 5 mesylphenyl ester, phenylazophenyl ester, phenyl thioester, p-nitrophenyl thioester, p-cresyl thioester, carboxymethyl thioester, pyranyl ester, pyridyl ester, piperidyl ester, 8-quinolyl thioester, etc.), or an ester with a N-hydroxy compound (e.g. N,N-dimethylhydroxylamine, 10 1-hydroxy-2-(1H)-pyridone, N-hydroxysuccinimide, N-hydroxybenzotriazole, N-hydroxyphthalimide, l-hydroxy-6-chloro-lH-benzotriazole, etc.); substituted or unsubstituted aryl isocyanate; substituted or unsubstituted aryl isothiocyanate, and the 15 These reactive derivatives can optionally be selected from them according to the kind of the compound (XXII) to be used.

The reaction is usually carried out in a conventional solvent such as water, acetone, dioxane, acetonitrile, chloroform, methylene chloride, ethylene chloride, tetrahydrofuran, ethyl acetate, N,N-dimethylformamide, pyridine or any other organic solvents which do not adversely influence the reaction. These conventional solvents may also be used in a mixture with water.

When the compound (XXII) is used in free acid form or its salt form in the reaction, the reaction is preferably carried out in the presence of a conventional condensing agent such as N,N'-dicyclohexylcarbodiimide;

N-cyclohexyl-N'-morpholinoethylcarbodiimide;
N-cyclohexyl-N'-(4-diethylaminocyclohexyl)carbodiimide;
N,N'-diethylcarbodiimide, N,N'-diisopropylcarbodiimide;
N-ethyl-N'-(3-dimethylaminopropyl)carbodiimide;
N,N-carbonylbis-(2-methylimidazole); pentamethyleneketeneN-cyclohexylimine, diphenylketene-N-cyclohexylimine;

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ethoxyacetylene; l-alkoxy-l-chloroethylene; trialkyl phosphite; ethyl polyphosphate; isopropyl polyphosphate; phosphorus oxychloride (phosphoryl chloride); phosphorus trichloride; thionyl chloride; oxalyl chloride; triphenylphosphine; 2-ethyl-7-hydroxybenzisoxasolium salt; 2-ethyl-5-(m-sulfophenyl)isoxazolium hydroxide intra-molecular salt; l-(p-chlorobenzenesulfonyloxy)-6-chloro-lH-benzotriazole; so-called Vilsmeier reagent prepared by the reaction of N,N-dimethylformamide with thionyl chloride, phosgene, phosphorus oxychloride, etc.; or the like.

The reaction may also be carried out in the presence of an inorganic or organic base such as an alkali metal bicarbonate, tri(lower)alkylamine, pyridine, N-(lower)alkylmorphorine, N,N-di(lower)alkylbenzylamine, or the like. The reaction temperature is not critical, and the reaction is usually carried out under cooling to heating.

### 20 Process (5)

The compound (If) or a salt thereof can be prepared by reacting the compound (XVIII) or a salt thereof with the compound (VI).

The reaction is usually carried out in a conventional solvent such as water, alcohol (e.g., methanol, ethanol, isopropyl alcohol, etc.), tetrahydrofuran, dioxane, chloroform, methylene chloride, dimethyl acetamide, N,N-dimethylformamide or any other organic solvent which does not adversely influence the reaction. Among these solvents, hydrophilic solvents may be used in a mixture with water.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

Suitable salts of the compounds (XVIII) and (If) can be referred to the ones as exemplified for the compound (I).

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### Process (6)

The compound (Ih) or a salt thereof can be prepared by subjecting the compound (Ig) or a salt thereof to reduction.

Reduction is carried out in a conventional manner including chemical reduction and catalytic reduction.

Suitable reducing agents to be used in chemical reduction are hydrides (e.g. hydrogen iodide, hydrogen sulfide, lithium aluminum hydride, sodium borohydride, etc.) or a combination of a metal (e.g. tin, zinc, iron, etc.) or metallic compound (e.g. chromium chloride, chromium acetate, etc.) and an organic or inorganic acid (e.g. formic acid, acetic acid, propionic acid, trifluoroacetic acid, p-toluenesulfonic acid, hydrochloric acid, hydrobromic acid, etc.).

Suitable catalysts to be used in catalytic reduction are conventional ones such as platinum catalysts (e.g. platinum plate, spongy platinum, platinum black, colloidal platinum, platinum oxide, platinum wire, etc.), palladium catalysts (e.g. spongy palladium, palladium black, palladium oxide, palladium on carbon, colloidal palladium, palladium on barium sulfate, palladium on barium carbonate, etc.), nickel catalysts (e.g. reduced nickel, nickel oxide, Raney nickel, etc.), cobalt catalysts (e.g. reduced cobalt, Raney cobalt, etc.), iron catalysts (e.g. reduced iron, Raney iron, etc.), copper catalysts (e.g. reduced copper, Raney copper, Ullman copper, etc.) and the like.

The reduction is usually carried out in a solvent

such as water, alcohol (e.g. methanol, ethanol, etc.),

N,N-dimethylformamide, tetrahydrofuran, a mixture thereof

or any other solvent which does not adversely affect the
reaction.

Additionally, in case that the above-mentioned acids to be used in chemical reduction are in liquid, they can

also be used as a solvent.

The reaction temperature of this reduction is not critical and the reaction is usually carried out under cooling to heating.

Suitable salts of the compounds (Ig) and (Ih) can be referred to the ones as exemplified for the compound (I).

### Process (7)

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The compound (Ij) or a salt thereof can be prepared by subjecting the compound (Ii) or a salt thereof to reduction as explained in Process (6).

Suitable salts of the compounds (Ii) and (Ij) can be referred to the ones as exemplified for the compound (I).

### 15 Process (8)

The compound (Ik) or a salt thereof can be prepared by reacting the compound (XXI) or a salt thereof with the compound (VII).

The reaction is usually carried out in a conventional solvent such as water, alcohols (e.g., methanol, ethanol, isopropyl alcohol, etc.), tetrahydrofuran, dioxane, chloroform, methylene chloride, dimethyl acetamide, xylene, 2-methoxyethanol, N,N-dimethylformamide or any other organic solvent which does not adversely influence the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

Suitable salts of the compounds (Ik) and (XXI) can be referred to the ones as exemplified for the compound (I).

### Process (9)

The compound (I1) or a salt thereof can be prepared by reacting the compound (VIII) or a salt thereof with the compound (IX) or a salt thereof.

35 This reaction is usually carried out in a

conventional solvent as exemplified in Process (8).

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

The reaction is usually carried out in the presence of an inorganic or an organic base such as an alkali metal hydroxide, an alkali metal hydrogencarbonate, alkali metal carbonate, alkali metal acetate, tri(lower)alkylamine, alkali metal methoxide, pyridine (e.g. pyridine, lutidine, picoline, dimethylaminopyridine, etc.),

N-(lower)alkylmorpholine, N,N-di(lower)alkylbenzylamine, N,N-di(lower)alkylaniline or the like. When the base and/or the starting compound are in liquid, they can be used also as a solvent.

Suitable salts of the compound (VIII) can be referred to the acid addition salts as exemplified for the compound (I).

Suitable salts of the compounds (IX) and (I1) can be referred to the ones as exemplified for the compound (I).

### 20 Process (10)

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The compound (In) or a salt thereof can be prepared by subjecting the compound (Im) or a salt thereof to reduction.

Reduction is carried out in a conventional manner, including chemical reduction and catalytic reduction.

Suitable reducing agents to be used in chemical reduction are hydrides (e.g. hydrogen iodide, hydrogen sulfide, lithium aluminum hydride, sodium borohydride, etc.), or a combination of a metal (e.g. tin, zinc, iron, etc.) or metallic compound (e.g. chromium chloride, chromium acetate, etc.) and an organic or inorganic acid (e.g. formic acid, acetic acid, propionic acid, trifluoroacetic acid, p-toluenesulfonic acid, hydrochloric acid, hydrobromic acid, etc.).

35 Suitable catalysts to be used in catalytic reduction

are conventional ones such as platinum catalysts (e.g. platinum plate, spongy platinum, platinum black, colloidal platinum, platinum oxide, platinum wire, etc.), palladium catalysts (e.g. spongy palladium, palladium black, palladium oxide, palladium on carbon, colloidal palladium 5 palladium on barium sulfate, palladium on barium carbonate, etc.), nickel catalysts (e.g. reduced nickel, nickel oxide, Raney nickel, etc.), cobalt catalysts (e.g. reduced cobalt, Raney cobalt, etc.), iron catalysts (e.g. reduced iron, Raney iron, etc.), copper catalysts (e.g. 10 reduced copper, Raney copper, Ullman copper, etc.) and the like. The reduction is usually carried out in a conventional solvent which does not adversely influence the reaction such as methanol, ethanol, propanol, N,N-dimethylformamide, tetrahydrofuran, or a mixture 15 thereof. Additionally, in case that the above-mentioned acids to be used in chemical reduction are in liquid, they

The reaction temperature of this reduction is not critical and the reaction is usually carried out under cooling to heating.

can also be used as a solvent.

Suitable salts of the compound (Im) can be referred to the ones as exemplified for the compound (I).

### 25 Process (11)

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The compound (Io) or a salt thereof can be prepared by subjecting the compound (In) or a salt thereof to oxidation reaction.

Oxidation is carried out in a conventional manner, which is capable of oxidizing a hydroxymethyl group to a formyl group, and suitable oxidizing reagent may be oxygen acid such as periodate (e.g. sodium periodate, etc.), peroxy acid such as peroxybenzoic acid (e.g. peroxybenzoic acid, m-chloroperoxybenzoic acid, etc.), magnese dioxide, and the like.

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The reaction is usually carried out in a conventional solvent as exemplified in Process (8), or any other organic solvent which does not adversely influence the reaction. Among these solvents, hydrophilic solvents may be used in a mixture with water.

The reaction temperature is not critical and the reaction is usually carried out under cooling to warming.

Suitable salts of the compounds (In) and (Io) can be referred to the acid addition salts as exemplified for the compound (I).

### Process (12)

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The compound (Ip) or a salt thereof can be prepared by reacting the compound (Io) or a salt thereof with the compound (X) or a salt thereof.

The reaction is usually carried out in a conventional solvent as exemplified in Process (8), or any other organic solvent which does not adversely influence the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

The reaction is usually carried out in the presence of an inorganic or an organic base such as an alkali metal hydroxide, an alkali metal hydrogencarbonate, alkali metal carbonate, alkali metal acetate, tri(lower)alkylamine, piperidine, piperazine, pyridine (e.g. pyridine, lutidine, picoline, dimethylaminopyridine, etc.),
N-(lower)alkylmorpholine, N,N-di(lower)alkylbenzylamine,
N,N-di(lower)alkylaniline or the like. When the base and/or the starting compound are in liquid, they can be used also as a solvent.

Suitable salts of the compound (X) can be referred to the base salts as exemplified for the compound (I).

Suitable salts of the compound (Ip) can be referred to the ones as exemplified for the compound (I).

### Process (13)

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The compound (Ir) or a salt thereof can be prepared by reacting the compound (Iq) or a salt thereof with the compound (XI) or a salt thereof.

The reaction is usually carried out in a conventional solvent as exemplified in Process (8), or any other organic solvent which does not adversely influence the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating. In case that the compound (XI) is liquid, it can be used as a solvent.

Suitable salts of the compounds (Ir) and (XI) can be referred to the ones as exemplified for the compound (I).

Process (14)

The compound (Is) or a salt thereof can be prepared by reacting the compound (Iq) or a salt thereof with the compound (XII) or a salt thereof.

The reaction is usually carried out in a conventional solvent as exemplified in Process (8), or any other organic solvent which does not adversely influence the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating. In case that the compound (XII) is liquid, it can be used as a solvent.

Suitable salts of the compound (Iq), (Is) and (XII) can be referred to the acid addition salts as exemplified for the compound (I).

### Process (15)

The compound (It) or a salt thereof can be prepared by reacting the compound (XIII) or a salt thereof with the compound (XIV) or a salt thereof.

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The reaction is usually carried out in a conventional solvent as exemplified in Process (8), or any other organic solvent which does not adversely influence the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to heating.

Suitable salts of the compounds (XIII), (XIV) and (It) can be referred to the ones as exemplified for the compound (I).

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### Process (16)

The compound (Iu) or a salt thereof can be prepared by reacting the compound (XV) or a salt thereof with the compound (XVI) and the compound (XVII) or a salt thereof.

The reaction is usually carried out in a conventional solvent such as alcohols (e.g. methanol, ethanol, ethylene glycol, etc.), chloroform, ether, tetrahydrofuran, benzene or any other organic solvent which does not adversely influence the reaction.

The reaction temperature is not critical and the reaction is usually carried out under cooling to warming.

The reaction is usually carried out in the presence of an acid.

Suitable acid may include an organic acid (e.g. formic acid, acetic acid, propionic acid, etc.) and an inorganic acid (e.g. hydrochloric acid, hydrobromic acid, sulfuric acid, etc.). When the acid is liquid, it can be used also as a solvent.

Suitable salts of the compounds (XV) (XVII) and (Iu) can be referred to the acid addition salts as exemplified for the compound (I).

### Process (17)

The compound (Iw) or a salt thereof can be prepared by subjecting the compound (Iv) or a salt thereof to

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reduction as explained in Process (10).

Suitable salts of the compounds (Iw) and (Iv) can be referred to the ones as exemplified for the compound (I).

#### 5 Process (a)

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The compound (III) or a salt thereof can be prepared by the compound (XIX) or a salt thereof with the compound (XX) or a salt thereof.

This reaction can be carried out in the presence or absence of a conventional solvent.

The reaction temperature is not critical and the reaction is usually carried out under cooling or at ambient temperature.

Suitable salts of the compound (XX) can be referred to the acid addition salts as exemplified for the compound (I).

Suitable salts of the compounds (XIX) and (III) can be referred to the ones as exemplified for the compound (I).

20 The new heterocyclic compounds (I) and a pharmaceutically acceptable salt thereof of the present invention possess strong antithrombotic activity inhibiting the activities against cyclooxygenase, thrombin, phosphodiesterase and the like, and/or inhibiting aggregation of platelet; vasodilating activity; 25 and anti-inflammatory activity; particularly antithrombotic activity, and therefore are useful as antithrombotic agent, vasodilating agent, and anti-inflammatory agent, particularly anti-thrombotic 30 agent.

> Accordingly, the new heterocyclic compounds (I) and a pharmaceutically acceptable salt thereof can be used for prophylactic and thrapeutic treatment of cerebral thrombosis, atrophic thrombosis; coronary thrombosis; creeping thrombosis; dilation thrombosis; jumping thrombosis; mural thrombosis; placental thrombosis;

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platelet thrombosis; posttraumatic arterial thrombosis; thrombostasis; compression thrombosis; peripheral vascular disorders such as chronic arterial occlusion; transient ischemic attack; myocardial infarction; cerebral infarction; reocclusion after percutaneous transluminal coronary angioplasty or percutaneous transluminal coronary recanalization; arteriosclerosis; cerebiral vasospam; disseminated intravascular coagulopathy; hypertension such as pulmonary hypertension; psoriasis; arthritis; nephritis; inflammatory bowel diseases; septic shock; dysmnesia; senile dementia; endotoxin shock; and the like.

And, these compounds are also useful for inhibition of thrombosis during extracorporeal circulation such as dialysis.

Further, these compounds are also expected to have antipyretic activity, analgesic activity, antiviral activity, antifungal activity, anti-allergic activity, 5-lipoxygenase inhibitory activity and the like.

The heterocyclic compounds (I) and a pharmaceutically acceptable salt thereof scarcely have side effect exerting a bad influence upon patients.

In order to show the utilities of the heterocyclic compounds (I) and a pharmaceutically acceptable salt thereof of the present invention, pharmacological test data of the representative compound of the heterocyclic compounds (I) are illustrated in the following.

The expressions of "Example 1-(1)", "Example 2-(1)" and "Example 7-(3)" "Example 17-(2)", "Example 21-(1)" and "Example 24-(1)" in the following tests mean the compounds prepared in Example 1-(1), 2-(1), 7-(3), 17-(2), 21-(1) and 24-(1) respectively.

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### Platelet aggregation ex vivo

### 1. Test method

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Male Hartley guinea-pigs weighing about 300 g were used after 24 hours fasting. Six hours after oral administration of the test compound or vehicle of test compound (control), blood was collected into a tube containing 0.1 vol. of 3.8% sodium citrate and platelet rich plasma (PRP) was prepared.

To the 250 ul of PRP, 5  $\mu$ l of arachidonic acid (final 50  $\mu$ M) was added as an aggregation inducer. Aggregation was measured by using an aggregometer (NKK HEMA-TRACER 1). The following result shows the relationship between the dose of the test compound and the percentage (%) of its inhibitory activity against the platelet aggregation responses.

### 2. Test result

Test compound Dose (mg/kg) Inhibition (%)

Example 2-(1) 1.0 100

Example 21-(1) 1.0 68.3

### 25 Relaxation effect on isolated rat aorta

### 1. Test method

Helical strip of rat thoracic aorta was suspended in the organ bath containing Tyrode solution gassed with 95% 0<sub>2</sub> - 5% CO<sub>2</sub> at 37°C under 0.5 g load. Contraction was induced by addition of KCl solution (final concentration was 30 mM). After the tonus reached plateau, drug solution (dissolved in dimethyl sulfoxide) was added cumulatively and finally 10<sup>-4</sup>M of papaverine was added to get maximum relaxation. Activities of the test compound were

expressed as  $ED_{50}$  values i.e. doses required to relax the isolated rat aorta by 50%.

### 2. Test result

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Test compounds	ED <sub>50</sub> (M)
Example 1-(1)	$4.3 \times 10^{-5}$
Example 2-(1)	$4.5 \times 10^{-5}$
Example 24-(1)	4.6 x 10 <sup>-6</sup>

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# Effect on malondialdehyde (MDA) production in rabbit platelets

### 1. Test method

Washed rabbit PRP (990  $\mu$ l) was preincubated with drug solution (dissolved in dimethyl sulfoxide) (10  $\mu$ l) at 37°C for 5 minutes. Then, 2.5 mM arachidonic acid solution (20  $\mu$ l) was added to the reaction mixture. After 3 minutes, thiobarbiturate reagent (1000  $\mu$ l) was added, and the reaction mixture was heated in a boiled water for 10 minutes. After centrifugation at 1500 g for 10 minutes, the absorbance of suparnatant was measured at 532 nm. This test was carried out to see inhibitory activity of the test compound against the activity of cyclooxygenase.

## 30 2. Test result

Test compound	concentration (M)	Inhibition (%)
Example 7-(3)	1.0 x 10 <sup>-7</sup>	73.5
Example 17-(2)	1.0 x 10 <sup>-7</sup>	80.1

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For therapeutic administration, the object compounds (I) of the present invention and pharmaceutically acceptable salts thereof are used in a form of the conventional pharmaceutical preparation in admixture with a conventional pharmaceutically acceptable carrier such as an organic or inorganic solid or liquid excipient which is suitable for oral, parenteral or external administration. The pharmaceutical preparation may be compounded in a solid form such as granule, capsule, tablet, dragee or suppository, or in a liquid form such as solution, suspension or emulsion for injection, ingestion, eye drops, etc. If needed, there may be included in the above preparation auxiliary substance such as stabilizing agent, wetting or emulsifying agent, buffer or any other commonly used additives.

The effective ingredient may usually be administered with a unit dose of 0.001 mg/kg to 500 mg/kg, preferably 0.01 mg/kg to 10 mg/kg, 1 to 4 times a day. However, the above dosage may be increased or decreased according to age, weight and conditions of the patient or the administering method.

The following preparations and examples are given only for the purpose of illustrating the present invention in more detail.

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### Preparation 1

To a solution of anisil (25 g) and sodium hydroxide (21.62 g) in methanol (1.5 l) was added 2,3-diaminopropanoic acid hydrobromide (36.52 g). The mixture was refluxed for 1 hour, and the cooled mixture was filtered. The filtrate was concentrated to a volume of 200 ml under reduced pressure and insolble material was collected by filtration. A mixture of this precipitate, water (100 ml) and ethyl acetate (50 ml) was adjusted to pH 10.5 with 3N-sodium hydroxide aqueous solution and the resultant mixture was acidified to pH 4.0 with 4N-hydrochloric acid. The precipitate was collected and washed with water to give 2,3-bis(4-methoxyphenyl)-pyrazine-5-carboxylic acid.

15 mp: 233°C (dec.)

IR (Nujol): 1690, 1600, 1510 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 2.80 (6H, s), 6.97, 7.46 (8H, ABq, J=9Hz), 9.13 (lH, s)

MASS (m/z): 336  $(M^{+})$ 

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### Preparation 2

A mixture of 2,3-bis(4-methoxyphenyl)pyrazine-5-carboxylic acid (0.5 g), 3-methyl-1-p-tolyltriazine (0.44 g) in tetrahydrofuran (15 ml) was stirred at 40°C for 1 hour. The reaction mixture was poured into a mixture of ethyl acetate and water and adjusted to pH \(\frac{1}{2}\) with 6N-hydrochloric acid. The separated organic layer was washed with brine and dried over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo, and the resulting precipitate was washed with isopropyl ether to give 5-methoxycarbonyl-2,3-bis(4-methoxyphenyl)pyrazine (0.36 g).

IR (Nujol): 1720, 1600, 1505 cm<sup>-1</sup>

NMR (CDCl<sub>3</sub>, δ): 3.82 (3H, s), 3.83 (3H, s),

4.04 (3H, s), 6.85 (4H, d, J=9Hz), 7.47 (2H, d,

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J=9Hz), 7.48 (2H, d, J=9Hz), 9.18 (1H, s) Mass (m/z): 350 (M<sup>+</sup>)

### Example 1

(1) To a mixture of 2-acetylamino-l-thioxoethylamine 5 (0.50 g) and ethanol was added hydrazine monohydrate (0.18 ml) at -70°C and the mixture was stirred at the same temperature for 30 minutes. To the reaction mixture containing 2-acetylamino-1-hydrazonoethylamine was added p-anisil (1.02 g) at -20°C, and an ethanol solution of 10 hydrogen chloride (3 drops) was added thereto. reaction mixture was stirred and refluxed for 20 minutes. After allowing to cool to room temperature, the mixture was poured into water, and extracted with ethyl acetate. The organic layer was washed with saturated aqueous sodium 15 bicarboante, water and brine, dried over magnesium sulfate and treated with active carbon. After filtration, the filtrate was evaporated in vacuo, the residue was subjected to column chromatography on silica gel (50 g) and eluted with a mixture of chloroform and methanol. The 20 fractions containing the object compound were combined and evaporated in vacuo, and the residue was triturated with isopropyl ether to give 3-(acetylaminomethyl)-5,6bis(4-methoxyphenyl)-1,2,4-triazine (0.20 g).

25 mp: 44-55°C IR (Nujol): 1650, 1600, 1490 cm<sup>-1</sup> NMR (CDCl<sub>3</sub>, δ): 2.13 (3H, s), 3.83 (3H, s), 3.85 (3H, s), 4.90 (2H, d, J=5Hz), 6.8-7.1 (5H, m), 7.55 (2H, d, J=9Hz), 7.58 (2H, d, J=9Hz) MASS (m/z): 364 (M<sup>+</sup>)

The following compounds were obtained according to a similar manner to that of Example 1-(1).

```
(2) 3-Ethoxycarbonyl-5,6-bis(4-methoxyphenyl)-1,2,4-
            triazine
            mp: 120-122°C
            IR (Nujol): 1740, 1650, 1595, 1570 cm<sup>-1</sup>
           NMR (CDCl<sub>3</sub>, \delta): 1.50 (3H, t, J=7Hz), 3.84 (3H, s),
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                 3.86 (3H, s), 4.6 (2H, q, J=7Hz), 6.88 (2H, d,
                 J=9Hz), 6.96 (2H, d, J=9Hz), 7.62 (2H, d,
                 J=9Hz), 7.69 (2H, d, J=9Hz)
           MASS (m/z): 365 (M^{+})
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      (3) 3-(2-tert-Butyloxycarbonylaminoethyl)-5,6-
           bis(4-methoxyphenyl)-1,2,4-triazine
           mp: 102-105°C
           IR (Nujol): 3320, 1680, 1610, 1490, 1250 cm<sup>-1</sup>
           NMR (DMSO-d_6, \delta): 1.33 (9H, s), 3.18 (2H, t, J=6Hz),
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                 3.48 (2H, q, J=6Hz), 3.78 (3H, s), 3.80 (3H, s),
                 6.95 (2H, d, J=9Hz), 6.99 (2H, d, J=9Hz),
                 7.43 (2H, d, J=9Hz), 7.52 (2H, d, J=9Hz)
           MASS (m/z): 436 (M^{+})
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      (4) 3-(1-tert-Butyloxycarbonylamino-1-methylethyl)-5,6-
           bis(4-methoxyphenyl)-1,2,4-triazine
           mp: 140°C
           IR (Nujol): 3250, 1700, 1600, 1490, 1300, 1250 cm<sup>-1</sup>
           NMR (DMSO-d_c, \delta): 1.31 (9H, s), 1.68 (6H, s),
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                 3.79 (3H, s), 3.80 (3H, s), 6.97 (2H, d, J=9Hz),
                 7.01 (2H, d, J=9Hz), 7.46 (2H, d, J=9Hz),
                 7.51 (2H, d, J=9Hz)
           MASS (m/z): 450 (M^{+})
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      (5) 3-(N,N-Dimethylaminomethyl)-5,6-bis(4-methoxyphenyl)-
           1,2,4-triazine
           IR (Neat): 2820, 1600, 1480, 1250 cm<sup>-1</sup>
           NMR (DMSO-d_6, \delta): 2.33 (6H, s), 3.79 (3H, s), 3.80
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                 (3H, s), 3.87 (2H, s), 6.95 (2H, d, J=8Hz), 6.99
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(2H, d, J=8Hz), 7.45 (2H, d, J=8Hz), 7.52 (2H, d, J=8Hz)

MASS (m/z): 350 (M<sup>+</sup>)
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5 (6) 5,6-Bis(4-methoxyphenyl)-3-{(4-methylpiperazin-l-yl)-carbonyl}-1,2,4-triazine hydrochloride

mp: 252-254°C

IR (Nujol): 3400, 2400, 1645, 1600, 1575 cm<sup>-1</sup>

### 10 Example 2

- (1) A mixture of 3-ethoxycarbonyl-5,6-bis(4-methoxyphenyl)-1,2,4-triazine (1.00g) and N-methylpiperazine (1.82 ml) was heated at 80-90°C for 4 hours and 40 minutes. After allowing to cool to room temperature, the mixture was poured into water, and
- extracted with ethyl acetate. The organic layer was washed with water and brine, dried over magnesium sulfate and treated with active carbon. After filtration, the filtrate was evaporated in vacuo, and the resulting matter
- was dissolved with diethyl ether, and to it was added an ethanol solution of hydrogen chloride. The resulting precipitate was collected by filtration, washed with ethanol and diethyl ether, and dried to give 5,6-bis(4-methoxyphenyl)-3-{(4-methylpiperazin-1-yl)-
- carbonyl}-1,2,4-triazine hydrochloride (0.76 g).

mp: 252-254°C

IR (Nujol): 3400, 2400, 1645, 1600, 1575 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 2.81 (3H, s), 2.80-4.80 (8H, m),

3.80 (3H, s), 3.81 (3H, s), 6.98 (2H, d, J=9Hz),

7.03 (2H, d, J=9Hz), 7.53 (4H, d, J=9Hz)

MASS (m/z): 418 [M<sup>+</sup> (419) of free compound-1]

The following compounds were obtained according to a similar manner to that of Example 2-(1).

```
(2) 3-{(4-(2-Hydroxyethyl)piperazin-l-yl)carbonyl}-5,6-
            bis(4-methoxyphenyl)-1,2,4-triazine hydrochloride
            mp: 141-147°C
            IR (Nujol): 1650, 1610 cm<sup>-1</sup>
            NMR (DMSO-d_6, \delta): 2.8-4.8 (12H, m), 3.80 (3H, s),
 5
                  3.81 (3H, s), 6.98 (2H, d, J=9Hz),
                  7.03 (2H, d, J=9Hz), 7.53 (4H, d, J=9Hz)
            MASS (m/z): 449 (M^+) of free compound)
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       (3) 3-(Morpholinocarbonyl)-5,6-bis(4-methoxyphenyl)-
            1,2,4-triazine
            mp: 65-70°C
            IR (Nujol): 1650, 1600, 1580 cm<sup>-1</sup>
            NMR (DMSO-d_6, \delta): 3.4-3.8 (8H, m), 3.79 (3H, s),
15
                 3.81 (3H, s), 6.96 (2H, d, J=9Hz),
                 7.03 (2H, d, J=9Hz), 7.53 (4H, d, J=9Hz)
            MASS (m/z): 406 (M^{+})
       (4) 3-(N,N-Dimethylaminocarbonyl)-5,6-bis(4-methoxy-
20
            phenyl)-1,2,4-triazine
            mp : 50-60°C
            IR (Nujol): 1650, 1600 cm<sup>-1</sup>
            NMR (DMSO-d_5, \delta): 2.98 (3H, s), 3.10 (3H, s),
                 3.79 (3H, s), 3.81 (3H, s), 6.96 (2H, d, J=9Hz),
25
                 7.01 (2H, d, J=9Hz), 7.53 (4H, d, J=9Hz)
            MASS (m/z): 364 (M^{+})
            3-[{2-(N,N-Dimethylamino)ethyl}carbamoyl]-5,6-
            bis(4-methoxyphenyl)-1,2,4-triazine
30
            mp: 137-140°C
            IR (Nujol): 1690, 1600, 1250 cm<sup>-1</sup>
           NMR (DMSO-d_6, \delta): 2.21 (6H, s), 2.48 (2H, t,
                 J=6Hz), 3.46 (2H, q, J=6Hz), 3.80 (3H, s), 3.81
                 (3H, s), 6.98 (2H, d, J=8Hz), 7.02 (2H, d,
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                 J=8Hz), 7.53 (2H, d, J=8Hz), 7.61 (2H, d,
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J=8Hz), 9.06 (1H, t, J=6Hz)
MASS (m/z): 407 (M<sup>+</sup>)
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- (6) 3-[(4-Benzylpiperazin-1-yl)carbonyl]-5,6-bis(4methoxyphenyl)-1,2,4-triazine
  mp: 126-129°C
  IR (Nujol): 1630,1600, 1480 cm<sup>-1</sup>
  NMR (CDCl<sub>3</sub>, δ): 2.4-2.7 (4H, m), 3.57 (2H, s),
  3.7-4.0 (4H, m), 3.83 (3H, s), 3.85 (3H, s),
  6.84 (2H, d, J=8.9Hz), 6.92 (2H, d, J=8.9Hz),
  7.2-7.45 (5H, m), 7.56 (2H, d, J=8.9Hz), 7.66
  (2H, d, J=8.9Hz)
- (7) 3-[(1-Benzylpiperidin-4-yl)carbamoyl]-5,6-bis(4methoxyphenyl)-1,2,4-triazine
  mp: 132-136°C
  IR (Nujol): 3380, 1690, 1600, 1520, 1490 cm<sup>-1</sup>
  NMR (CDCl<sub>3</sub>, δ): 1.55-2.4 (6H, m), 2.75-3.0 (2H, m),
  3.54 (2H, s), 3.84 (3H, s), 3.86 (3H, s),
  4.03-4.18 (1H, m), 6.86 (2H, d, J=8.9Hz), 6.93
  (2H, d, J=8.9Hz), 7.18-7.5 (5H, m), 7.59 (2H, d,
  J=8.9Hz), 7.69 (2H, d, J=8.9Hz), 8.01 (1H, d,
  J=8.2Hz)
  MASS (m/z): 509 (M<sup>+</sup>)

Example 3

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A mixture of 3-(acetylaminomethyl)-5,6-bis(4-methoxy-phenyl)-1,2,4-triazine (0.79 g) and concentrated hydrochloric acid (5 ml) was stirred and refluxed for 2 hours. After allowing to cool to room temperature, the mixture was poured into water. Then, an aqueous solution of sodium bicarbonate was added thereto to adjust to pH 10, and extracted with ethyl acetate. The organic layer was washed with water and brine, dried over magnesium sulfate and treated with active carbon. After filtration,

the filtrate was evaporated in vacuo, and the resulting matter was dissolved with ethanol and to it was added an ethanol solution of hydrogen chloride. To the resulting mixture was added diethyl ether and it was triturated to give powder of 3-(aminomethyl)-5,6-bis(4-methoxyphenyl)-1,2,4-triazine hydrochloride (0.18 g).

mp: 163-173°C (decomp.)
IR (Nujol): 1600, 1580, 1520 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 3.80 (3H, s), 3.81 (3H, s),
4.52 (2H, br s), 6.98 (2H, d, J=9Hz),
7.03 (2H, d, J=9Hz), 7.49 (2H, d, J=9Hz),
7.63 (2H, d, J=9Hz), 9.00 (3H, br s)

MASS (m/z): 322 (M<sup>+</sup> of free compound)

### 15 Example 4

A mixture of 3-(aminomethyl)-5,6bis(4-methoxyphenyl)-1,2,4-triazine hydrochloride (1.00 g), chloroform and an aqueous solution of sodium bicarbonate was stirred at ambient temperature for 30 20 minutes, and the separated organic layer was dried over magnesium sulfate and treated with active carbon. After filtration, the filtrate was evaporated in vacuo. resulting matter was dissolved with N,N-dimethylformamide (10 ml), and to the mixture were added 6-carboxyl-3-oxo-25 2,3,4,5-tetrahydropyridazine (0.40 g), l-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (0.53 g) and triethylamine (0.39 ml). resulting mixture was stirred at 100°C for 15 hours. After allowing to cool to room temperature, the mixture 30 was poured into water, and extracted with ethyl acetate. The separated organic layer was washed with water, diluted aqueous hydrochloric acid, a saturated aqueous solution of sodium bicarbonate, water and brine, dried over magnesium sulfate and treated with active carbon. After filtration, 35 the filtrate was evaporated in vacuo, the residue was

subjected to column chromatography on silica gel (25 g) and eluted with a mixture of chloroform and methanol. The fractions containing the object compound were combined and evaporated in vacuo, and the residue was triturated with ethanol and diethyl ether to give 5,6-bis(4-methoxyphenyl)-3-{(3-oxo-2,3,4,5-tetrahydro-pyridazin-6-yl)carbonylaminomethyl}-1,2,4-triazine (0.07 g).

mp: 183-187°C

IR (Nujol): 3400, 1700, 1670, 1660, 1640, 1610,

1580 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 2.47 (2H, t, J=8.2Hz), 2.78 (2H,

t, J=8.2Hz), 3.78 (3H, s), 3.80 (3H, s),

4.78 (2H, d, J=5.8Hz), 6.95 (2H, d, J=8.3Hz),

7.00 (2H, d, J=8.3Hz), 7.45 (2H, d, J=8.3Hz),

7.49 (2H, d, J=8.3Hz), 8.75 (1H, t, J=5.8Hz)

MASS (m/z): 446 (M<sup>+</sup>)

### Example 5

(1) A mixture of 5,6-bis(4-methoxyphenyl)-3-[2-(tert-butyloxycarbonylamino)ethyl]-1,2,4-triazine (1.00 g) and dichloromethane (10 ml) was stirred at 2°C. To the reaction mixture was added 4N-hydrogen chloride-1,4-dioxane (10 ml), and the mixture was stirred at ambient temperature for 1 hour. The resulting matter was evaporated in vacuo. To the residue was added isopropyl ether (50 ml) and it was stirred at 2°C for 3 hours. The resulting precipitate was collected by filtration and washed with isopropyl ether to give 3-(2-aminoethyl)-5,6-bis(4-methoxyphenyl)-1,2,4-triazine hydrochloride (0.73 g).

mp: 98°C (decomp.)

IR (Nujol): 3500-3300, 1600, 1510, 1310, 1260 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>,  $\delta$ ): 3.17-3.70 (4H, m), 3.79 (3H, s),

3.80 (3H, s), 6.97 (3H, d, J=9Hz),

```
7.01 (3H, d, J=9Hz), 7.46 (3H, d, J=9Hz), 7.53 (3H, d, J=9Hz), 8.01 (2H, br s)

MASS: m/z 336 (M<sup>+</sup> of free compound)
```

- 5 The following compound was obtained according to a similar manner to that of Example 5-(1).
  - (2) 3-(1-Amino-1-methylethyl)-5,6-bis(4-methoxyphenyl)1,2,4-triazine hydrochloride

10 mp: 245-247°C

IR (Nujol): 3600-3300, 1600, 1490, 1300 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 1.79 (6H, s), 3.80 (3H, s),

3.81 (3H, s), 6.99 (2H, d, J=9Hz),

7.03 (2H, d, J=9Hz), 7.50 (2H, d, J=9Hz),

7.65 (2H, d, J=9Hz), 8.93 (3H, br s)

MASS (m/z): 350 (M<sup>+</sup>)

### Example 6

To a solution of 3-(1-amino-1-methylethyl)-5,6-20 bis(4-methoxyphenyl)-1,2,4-triazine hydrochloride (1.30 g), 3-oxo-6-carboxy-2,3,4,5-tetrahydropyridazine (0.48 g) and 1-hydroxybenzotriazole hydrate (0.52 g) in N,N-dimethylformamide (30 ml) was added 1-(3dimethylaminopropyl)-3-ethylcarbodiimide (0.62 ml) under 25 stirring at ambient temperature and the resulting mixture was stirred for one hour at same condition. The reaction mixture was poured into water and extracted with ethyl acetate. The extract was washed with a saturated aqueous solution of sodium bicarbonate, water and brine and dried 30 over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo. The resulting precipitate was collected by filtration, and washed with diethyl ether to give  $5,6-bis(4-methoxyphenyl)-3-[1-{(3-oxo-2,3,4,5$ tetrahydropyridazin-6-yl)carbonylamino}-1-methylethyl]-35 1,2,4-triazine (1.33 g).

```
mp: 221-222°C
           IR (Nujol): 3350, 3250, 1700, 1660, 1600, 1500,
                          1250 cm<sup>-1</sup>
           NMR (DMSO-d_6, \delta): 1.83 (6H, s), 2.42 (2H, t, J=8Hz),
                 2.72 (2H, t, J=8Hz), 3.79 (3H, s), 3.80 (3H, s),
5
                 6.97 (2H, d, J=9Hz), 7.02 (2H, d, J=9Hz),
                 7.48 (2H, d, J=9Hz), 7.53 (2H, d, J=9Hz),
                 8.72 (lH, s), ll.20 (lH, s)
           MASS (m/z): 474 (M^+)
10
            The following compound was obtained according to a
       similar manner to that of Example 6-(1).
            5,6-Bis(4-methoxyphenyl)-3-[2-{(3-oxo-2,3,4,5-
       tetrahydropyridazin-6-yl)carbonylamino}ethyl]-1,2,4-
15
       triazine
            mp: 85°C (decomp.)
            IR (Nujol): 3500-3200, 1660, 1600, 1250 cm<sup>-1</sup>
            NMR (DMSO-d_6, \delta): 2.36 (2H, t, J=8Hz),
                 2.70 (2H, t, J=8Hz), 3.30-3.36 (2H, m),
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                 3.78 (3H, s), 3.80 (3H, s), 6.91 (2H, d, J=9Hz),
                 7.00 (2H, d, J=9Hz), 7.45 (2H, d, J=9Hz),
                 7.50 (2H, d, J=9Hz), 8.33 (1H, t, J=6Hz),
                 11.10 (lH, s)
            MASS (m/z): 460 (M^{+})
25
       Example 7
            A mixture of 3-(aminomethyl)-5,6-
       bis(4-methoxyphenyl)-1,2,4-triazine hydrochloride (0.80
       g), chloroform and an aqueous solution of sodium
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       bicarbonate was stirred at ambient temperature for 30
       minutes, and the separated organic layer was dried over
       magnesium sulfate and treated with active carbon. After
       filtration, the filtrate was evaporated in vacuo.
       resulting matter was added a mixture of tetrahydrofuran
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(20 ml) and methanol (7 ml), and to the mixture was added isopropyl isocyanate (0.26 ml). The reaction mixture was stirred at ambient temperature for 3 hours and 50 minutes. The mixture was evaporated in vacuo, and the resulting matter was triturated with isopropyl ether and diethyl ether to give 3-(N'-isopropylureidomethyl)-5,6-bis(4-methoxyphenyl)-1,2,4-triazine (0.75 g).

mp: 80-83°C

IR (Nujol): 3300, 1630, 1600, 1560, 1490 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 1.06 (6H, d, J=6.5Hz),

3.70 (1H, m), 3.78 (3H, s), 3.80 (3H, s),

4.64 (2H, d, J=5.8Hz), 6.14 (1H, d, J=7.7Hz),

6.45 (1H, t, J=5.8Hz), 6.95 (2H, d, J=8.8Hz),

7.00 (2H, d, J=8.8Hz), 7.45 (2H, d, J=8.8Hz),

7.52 (2H, d, J=8.8Hz)

MASS (m/z): 407 (M<sup>+</sup>)

The following compounds were obtained according to a similar manner to that of Example 7-(1).

(2) 3-(N'-Ethylureidomethyl)-5,6-bis(4-methoxyphenyl)1,2,4-triazine
mp: 69-74°C

IR (Nujol): 3300, 1640, 1600, 1560 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 1.03 (3H, t, J=7Hz),
3.05 (2H, q, J=7Hz), 3.78 (3H, s), 3.80 (3H, s),
4.64 (2H, d, J=5.9Hz), 6.23 (1H, t, J=5.5Hz),
6.56 (1H, t, J=5.5Hz), 6.95 (2H, d, J=9Hz),
7.00 (2H, d, J=9Hz), 7.45 (2H, d, J=9Hz),
7.52 (2H, d, J=9Hz)

MASS (m/z): 393 (M<sup>+</sup>)

(3) 3-[2-(N'-Isopropylureido)ethyl]-5,6-bis(4methoxyphenyl)-1,2,4-triazine
mp: 157-160°C

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IR (Nujol): 3400-3200, 1640, 1620, 1570, 1180 cm<sup>-1</sup>
          NMR (DMSO-d_6, \delta): 0.98 (6H, d, J=6.5Hz), 3.20 (2H,
               t, J=7Hz), 3.51-3.72 (3H, m), 3.79 (3H, s),
               3.80 (3H, s), 5.73 (1H, d, J=7.5Hz),
               5.85 (lH, t, J=6Hz), 6.96 (2H, d, J=9Hz),
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               6.99 (2H, d, J=9Hz), 7.45 (2H, d, J=9Hz),
               7.52 (2H, d, J=9Hz)
          MASS (m/z) : 421 (M^+)
     Example 8
     (1) A solution of 5,6-bis(4-methoxyphenyl)-3-(pyridin-4-
```

y1)-1,2,4-triazine (0.50 g) and methyl iodide (0.75 ml) in a mixture of chloroform and methanol (9:1) (20 ml) was allowed to stand at ambient temperature for 1 day. The reaction mixture containing 4-[5,6-bis(4-methoxyphenyl)-15 triazin-3-yl]-1-methylpyridinium iodide was evaporated in vacuo and the residue was dissolved in a mixture of methanol (30 ml) and water (10 ml). To the resultant solution was added portionwise sodium borohydride (0.10 g) under stirring at 5 to 10°C. The reaction mixture was 20 stirred for 2.5 hours at the same temperature. Water (100 ml) was added to the reaction mixture and the resulting precipitate was collected by filtration. The precipitate was dissolved in chloroform, washed with brine and dried over magnesium sulfate. The separated organic layer was 25 evaporated in vacuo, and the residue was washed with diethyl ether to give 5,6-bis(4-methoxyphenyl)-3-(l-methyl-1,2,3,6-tetrahydropyridin-4-yl)-1,2,4-triazine (0.32 g).

mp : 163-165°C 30 IR (Nujol): 1600, 1300, 1250 cm<sup>-1</sup> NMR (DMSO- $d_6$ ,  $\delta$ ): 2.33 (3H, s), 2.64 (2H, m), 2.76 (2H, m), 3.17 (2H, m), 3.78 (3H, s), 3.80 (3H, s), 6.95 (2H, d, J=9Hz),7.00 (2H, d, J=9Hz), 7.33 (1H, m), 35

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7.47 (2H, d, J=9Hz), 7.54 (2H, d, J=9Hz) MASS (m/z): 388  $(M^+)$ 

The following compound was obtained according to a similar manner to that of Example 8-(1).

### Example 9

(1) A mixture of 3-ethoxycarbonyl-5,6-bis(4-methoxyphenyl)-1,2,4-triazine (10 g) and 2,5-norbornadiene
20 (5.55 g) in the xylene (400 ml) was refluxed under
stirring for 24 hours. The reaction mixture was
evaporated in vacuo, and the resulting precipitate was
washed with diethyl ether to give
6-ethoxycarbonyl-2,3-bis(4-methoxyphenyl)pyridine
25 (6.47 g).

mp: 108-109°C

IR (Nujol): 1700, 1600, 1505 cm<sup>-1</sup>

NMR (CDCl<sub>3</sub>, δ): 1.45 (3H, t, J=7Hz), 3.79 (3H, s),

3.81 (3H, s), 4.48 (2H, q, J=7Hz), 6.78 (2H, d,

J=8Hz), 6.84 (2H, d, J=8Hz), 7.12 (2H, d,

J=8Hz), 7.37 (2H, d, J=8Hz), 7.78 (1H, d,

J=8Hz), 8.05 (1H, d, J=8Hz)

MASS (m/z): 363 (M<sup>+</sup>)

The following compounds were obtained according to a

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similar manner to that of Example 9-(1).

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(2) 6-Acetylaminomethyl-2,3-bis(4-methoxyphenyl)pyridine IR (Film): 1720 cm<sup>-1</sup>

NMR (CDCl<sub>3</sub>, δ): 2.08 (3H, s), 3.80 (3H, s), 3.81 (3H, s), 4.63 (2H, d, J=5Hz), 6.7-7.0 (5H, m), 7.07 (2H, d, J=8Hz), 7.2-7.4 (3H, m), 7.64 (1H, d, J=8Hz)

MASS (m/z): 362 (M<sup>+</sup>)
```

(3) 6-(Pyridin-4-yl)-2,3-bis(4-methoxyphenyl)pyridine
mp: 175-178°C
IR (Nujol): 1610, 1600, 1510, 1250 cm<sup>-1</sup>
NMR (DMSO-d<sub>6</sub>, δ): 3.77 (6H, s), 6.90 (2H, d, J=8Hz),
6.94 (2H, d, J=8Hz), 7.19 (2H, d, J=8Hz),
7.37 (2H, d, J=8Hz), 7.93 (1H, br s),

8.15 (lH, br s), 8.74 (2H, br s) MASS (m/z): 368  $(M^+)$ 

- 20 (4) 2,3-Bis(4-methoxyphenyl)-6-(N,N-dimethylaminomethyl)pyridine
  mp: 103-108°C (dec.)
  IR (Nujol): 1610, 1510, 1250 cm<sup>-1</sup>
- NMR (DMSO-d<sub>6</sub>, δ): 2.82 (3H, s), 2.83 (3H, s), 3.74
  (3H, s), 3.76 (3H, s), 4.46 (2H, s), 6.84 (2H,
  d, J=8Hz), 6.90 (2H, d, J=8Hz), 7.13 (2H, ,
  J=8Hz), 7.32 (2H, d, J=8Hz), 7.63 (1H, d,
  J=8Hz), 7.85 (2H, d, J=8Hz)
- 30 Example 10
  - (1) A mixture of 6-ethoxycarbonyl-2,3bis(4-methoxyphenyl)pyridine (1.03 g) and N-methylpiperazine (1.87 ml) was stirred and refluxed for 26.5 hours. After allowing to cool to room temperature,
- 35 the reaction mixture was poured into water and ethyl

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acetate. The separated organic layer was washed with water, saturated aqueous sodium bicarbonate, water and brine, and dried over magnesium sulfate and treated with active carbon.

After filtration, the filtrate was evaporated in vacuo, and the resulting residue was dissolved with diethyl ether, and to it was added an ethanol solution of hydrogen chloride. The resulting precipitate was collected by filtration and washed with ethanol and diethyl ether to give 2,3-bis(4-methoxyphenyl)-6-[(4-methyl-piperazin-l-yl)carbonyl]pyridine dihydrochloride (0.25 g).

mp: 213-216°C

IR (Nujol): 3400 (br), 2400 (br), 1980 (br), 1650, 1610, 1510 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 2.77 (3H, s), 2.8-4.8 (8H, m), 3.75 (3H, s), 3.76 (3H, s), 6.85 (2H, d, J=9Hz), 6.91 (2H, d, J=9Hz), 7.15 (2H, d, J=9Hz),

7.26 (2H, d, J=9Hz), 7.68 (1H, d, J=9Hz),

7.92 (lH, d, J=9Hz)

MASS (m/z): 415  $[M^+(417)$  of free compound -2]

The following compounds were obtained according to a similar manner to that of Example 10-(1).

(2) 2,3-Bis(4-methoxyphenyl)-6-[{2-(N,N-dimethylamino)ethyl}carbamoyl]pyridine dihydrochloride

mp: 68-71°C

IR (Nujol): 1600, 1510, 1300, 1250 cm<sup>-1</sup>

30 NMR (DMSO- $d_6$ ,  $\delta$ ): 2.80 (3H, s), 2.83 (3H, s),

3.28 (2H, q, J=6Hz), 3.73 (2H, t, J=6Hz), 3.75 (3H, s), 3.76 (3H, s), 6.86 (2H, d, J=8Hz),

6.91 (2H, d, J=8Hz), 7.15 (2H, d, J=8Hz),

7.42 (2H, d, J=8Hz), 7.92 (2H, d, J=8Hz),

35 8.00 (2H, d, J=8Hz), 9.02 (1H, t, J=6Hz),

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10.58 (lH, m)
MASS (m/z): 405  $(M^+)$  of free compound)

(3) 2,3-Bis(4-methoxyphenyl)-6-[(4-benzylpiperazin-1-yl)-carbamoyl]pyridine

mp: 126-132°C

IR (Nujol): 1610, 1500 cm<sup>-1</sup>

NMR (CDCl<sub>3</sub>,  $\delta$ ): 2.4-2.7 (4H, m), 3.56 (2H, s),

3.7-3.95 (4H, m), 3.79 (3H, s), 3.81 (3H, s),

6.77 (2H, d, J=8.9Hz), 6.83 (2H, d, J=8.9Hz),

7.11 (2H, d, J=8.9Hz), 7.2-7.45 (7H, m),

7.63 (lH, d, J=7.9Hz), 7.75 (lH, d, J=7.9Hz)

MASS (m/z): 493  $(M^{+})$ 

### 15 Example 11

- (1) To a mixture of lithium aluminum hydride (0.24 g) and tetrahydrofuran (20 ml) was added a solution of 6-ethoxycarbonyl-2,3-bis(4-methoxyphenyl)pyridine (2.1 g) in tetrahydrofuran (2.1 ml) with stirring and ice-cooling,
- and the mixture was stirred for 1 hour at the same temperature. To the reaction mixture were added ethyl acetate (20 ml) and water (20 ml) very carefully. The organic layer was washed with brine and dried over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo to give 6-hydroxymethyl-2,3-

evaporated in vacuo to give 6-hydroxymethy1-2,3-bis(4-methoxyphenyl)pyridine (2.04 g).

NMR (CDCl<sub>3</sub>,  $\delta$ ): 3.79 (3H, s), 3.81 (3H, s),

4.82 (2H, s), 6.79 (2H, d, J=9Hz),

6.84 (2H, d, J=9Hz), 7.0-7.4 (5H, m),

7.67 (lh, d, J=8Hz)

Mass (m/z): 321  $(M^{+})$ 

(2) A mixture of 6-hydroxymethyl-2,3bis(4-methoxyphenyl)pyridine (2 g) and activated manganese dioxide (6 g) in the chloroform (20 ml) was

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stirred at ambient temperature for 5.5 hours. After filtration, the filtrate was evaporated in vacuo, and the resulting precipitate was washed with isopropyl ether to give 2,3-bis(4-methoxyphenyl)-6-pyridinecarbaldehyde (1.61 g).

MMR (CDCl<sub>3</sub>, δ): 3.81 (3H, s), 3.82 (3H, s), 6.81 (2H, d, J=9Hz), 6.84 (2H, d, J=9Hz), 7.12 (2H, d, J=9Hz), 7.37 (2H, d, J=9Hz), 7.81 (1H, d, J=8Hz), 7.95 (1H, d, J=8Hz)

10 MASS (m/z): 319  $(M^+)$ 

### Example 12

The following compound was obtained by reacting 5-methoxycarbonyl-2,3-bis(4-methoxyphenyl)pyrazine according to a similar manner to that of Example 11.

2,3-Bis(4-methoxyphenyl)-6-pyrazinecarbaldehyde
IR (Nujol): 1700, 1600, 1505 cm<sup>-1</sup>
NMR (CDCl<sub>3</sub>, δ): 3.83 (3H, s), 3.84 (3H, s),
6.8-7.0 (4H, m), 7.44-7.6 (4H, m),
9.07 (1H, s), 10.22 (1H, s)
MASS (m/z): 320 (M<sup>+</sup>)

### Example 13

25 (1) A mixture of 2,3-bis(4-methoxyphenyl)-6pyridinecarbaldehyde (0.6 g), malonic acid (0.98 g),
piperidine (0.1 ml) and pyridine (6 ml) was refluxed under
stirring for 1 hour and 20 minutes. After allowing to
cool to room temperature, the mixture was poured into
30 water and ethyl acetate, and adjusted to pH =10.5 with an
aqueous solution of 4N sodium hydroxide. The separated
aqueous layer was washed with ethyl acetate, and the
aqueous layer was adjusted to pH =4 with 6N hydrochloric
acid and extracted with ethyl acetate. The organic layer
35 was washed with brine and dried over magnesium sulfate.

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After filtration, the filtrate was evaporated in vacuo to give 3-[2,3-bis(4-methoxyphenyl)pyridin-6-yl]-(E)propenoic acid (0.35 g). mp: 160-170°C IR (Nujol): 1680, 1630, 1600, 1500 cm<sup>-1</sup> 5 NMR (DMSO- $d_6$ ,  $\delta$ ): 3.75 (6H, s), 6.75-6.97 (4H, m), 6.85 (lH, d, J=15.6Hz), 6.99-7.35 (4H, m), 7.67 (lH, d, J=16Hz), 7.69 (lH, d, J=8Hz), 7.8 (lH, d, J=8Hz) MASS (m/z) : 361  $(M^+)$ 10 The following compound was obtained according to a similar manner to that of Example 13-(1). (2) 3-[2,3-Bis(4-methoxyphenyl)pyrazin-5-yl]-(E)-15 propenoic acid 238-240°C mp: IR (Nujol): 1680, 1630, 1600, 1500 cm<sup>-1</sup> NMR (CDCl<sub>3</sub> + DMSO-d<sub>6</sub>,  $\delta$ ) : 3.82 (3H, s), 3.83 (3H, 20 s), 6.85-6.93 (4H, m), 7.07 (1H, d, J=15.6Hz), 7.35-7.54 (4H, m), 7.75 (1H, d, J=15.6Hz), 8.56 (lH, s) MASS (m/z): 362  $(M^{+})$ 25 Example 14 (1) A mixture of 3-[2,3-bis(4-methoxyphenyl)pyridin-6-yl]-(E)-propenoic acid (0.8 g) and 10% palladium-carbon (0.16 g) in methanol (8 ml) was stirred under hydrogen at low pressure at ambient temperature for 5 hours. 30 insoluble material was filtered and washed with a mixture of chloroform and methanol, and the filtrate was

evaporated in vacuo. The residue was dissolved in a mixture of ethyl acetate and water and the resultant

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solution was adjusted to pH =4 with 6N-hydrochloric acid.

The organic layer was washed with brine and dried over

magnesium sulfate. The solvent was evaporated in vacuo and the residue was subjected to column chromatography on silica gel and eluted with a mixture of chloroform and methanol (10:1). The eluted solution was evaporated in vacuo and the resulting precipitate was washed with diethyl ether to give 3-[2,3-bis(4-methoxyphenyl)pyridin-6-yl]propanoic acid (495 mg).

mp: 146-147°C

IR (Nujol): 1710, 1600, 1510 cm<sup>-1</sup>

NMR (CDCl<sub>3</sub>, δ): 2.8-2.95 (2H, m), 3.15-3.29 (2H, m),

3.8 (3H, s), 3.81 (3H, s), 6.75-7.33 (9H, m),

7.76 (1H, d, J=8Hz)

MASS (m/z): 363 (M<sup>+</sup>)

- The following compound was obtained according to a similar manner to that of Example 14-(1).
  - (2) 3-[2,3-Bis(4-methoxyphenyl)pyrazin-5-yl]propanoic acid

20 mp: 120-128°C

IR (Nujol): 1700, 1600, 1570, 1500 cm<sup>-1</sup>

NMR (CDCl<sub>3</sub>, δ): 2.8-3.0 (2H, m), 3.1-3.3 (2H, m),

3.8 (3H, s), 3.81 (3H, s), 6.83 (2H, d, J=9Hz),

6.84 (2H, d, J=9Hz), 7.37 (2H, d, J=9Hz),

7.38 (2H, d, J=9Hz), 8.46 (1H, s)

MASS (m/z): 364 (M<sup>+</sup>)

### Example 15

(1) A mixture of 6-acetylaminomethyl-2,3-bis(4methoxyphenyl)pyridine (1.64 g) and concentrated
hydrochloric acid (3.3 ml) was stirred and refluxed for
l.5 hours. The reaction mixture was poured into a mixture
of ethyl acetate and ice-water and adjusted to pH =10 with
4N-sodium hydroxide. The separated organic layer was
washed with brine and dried over magnesium sulfate. After

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filtration, the filtrate was evaporated in vacuo, and the resulting residue was dissolved in ethanol and to it was added an ethanol solution of hydrogen chloride. To the resulting mixture was added diethyl ether and it was triturated to give 6-aminomethyl-2,3-bis(4-methoxyphenyl)pyridine dihydrochloride (1.13 g).

mp: 150°C (dec.)
IR (Nujol): 3400 (br), 1600, 1500 cm<sup>-1</sup>
NMR (CDCl<sub>3</sub>, δ): 3.71 (3H, s), 3.76 (3H, s),
5.2 (2H, br s), 6.75 (2H, d, J=9Hz),
6.81 (2H, d, J=9Hz), 7.01 (2H, d, J=9Hz),
7.45 (2H, d, J=9Hz), 8.31 (1H, d, J=8Hz),
8.54 (1H, d, J=8Hz), 9.49 (2H, br s)
MASS (m/z): 320 (M<sup>+</sup> of free compound)

The following compound was obtained according to a similar manner to that of Example 15-(1).

(2) 2-Aminomethyl-4,5-bis(4-methoxyphenyl)pyrimidine hydrochloride IR (Nujol): 1605, 1585, 1570, 1510 cm<sup>-1</sup> NMR (DMSO-d<sub>6</sub>, δ): 3.78 (6H, s), 4.33 (2H, d, J=6Hz), 4.82 (2H, s), 6.90 (2H, d, J=9Hz), 6.98 (2H, d, J=9Hz), 7.21 (2H, d, J=9Hz), 7.50 (2H, d, J=9Hz), 8.76 (1H, s)

### Example 16

(1) A mixture of 6-aminomethyl-2,3-bis(4-methoxyphenyl)pyridine dihydrochloride (0.2 g), 6-carboxy-3-oxo-2,3,4,5tetrahydropyridazine (72 mg), N-hydroxybenzotriazole (69
mg), 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide
hydrochloride (97 mg) and triethylamine (0.15 g) in
tetrahydrofuran (4 ml) was stirred at ambient temperature
for 15 hours. The reaction mixture was poured into water

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and the resulting precipitate was washed with water and tetrahydrofuran to give 2,3-bis(4-methoxyphenyl)-6-[(3-oxo-2,3,4,5-tetrahydropyridazin-6-yl)-carbonylaminomethyl]pyridine (126 mg).
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5 mp : 234-235°C

IR (Nujol) : 3380, 1670, 1655, 1620, 1500 cm<sup>-1</sup>

NMR (CDCl<sub>3</sub>, δ) : 2.55 (2H, t, J=8Hz),

2.98 (2H, t, J=8Hz), 3.80 (3H, s), 3.81 (3H, s),

4.70 (2H, d, J=5Hz), 6.73-6.87 (4H, m),

7.08 (2H, d, J=9Hz), 7.2-7.36 (4H, m),

7.64 (1H, d, J=8Hz), 8.1 (1H, br s),

8.63 (1H, s)

MASS (m/z) : 444 (M<sup>+</sup>)
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The following compound was obtained according to a similar manner to that of Example 16-(1)

Example 17

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(1) A mixture of 6-aminomethyl-2,3-bis(4-methoxyphenyl)pyridine dihydrochloride (0.3 g), dichloromethane and an
aqueous solution of sodium hydrogencarbonate was stirred
at ambient temperature for 30 minutes, and the separated

organic layer was dried over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo. To the resulting residue was added a mixture of tetrahydrofuran (6 ml) and methanol (2 ml), and then added isopropyl isocyanate (0.1 ml). The reaction mixture was stirred at ambient temperature for 14 hours. The mixture was evaporated in vacuo, and the resulting residue was triturated with isopropyl ether to give 6-(3-isopropylureidomethyl)-2,3-bis(4-methoxyphenyl)-pyridine (0.24 g).

mp: 120°C (dec.)
IR (Nujol): 3300, 1610, 1565, 1500 cm<sup>-1</sup>
NMR (CDCl<sub>3</sub>, δ): 1.13 (6H, d, J=6.5Hz), 3.80 (3H, s),
3.81 (3H, s), 3.90 (1H, m), 4.52 (2H, d, J=5Hz),
4.71 (1H, br), 5.54 (1H, br), 6.73-6.86 (4H, m),
7.0-7.4 (7H, m), 7.64 (1H, d, J=8Hz)

MASS (m/z): 405 (M<sup>+</sup>)

The following compound was obtained according to a similar manner to that of Example 17-(1).

(2) 4,5-Bis(4-methoxyphenyl)-2-(3-isopropylureidomethyl)-pyrimidine

NMR (DMSO-d<sub>6</sub>, δ): 1.05 (6H, d, J=7Hz), 3.69 (lH, m),
3.72 (6H, s), 4.45 (2H, d, J=6Hz), 6.15 (lH, d,
J=8Hz), 6.28 (lH, t, J=6Hz), 6.88 (2H, d,
J=9Hz), 6.91 (2H, d, J=9Hz), 7.18 (2H, d,
J=9Hz), 7.40 (2H, d, J=9Hz), 8.64 (lH, s)

MASS (m/z): 406 (M<sup>+</sup>)

Example 18

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(1) A mixture of 1,2-bis(4-methoxyphenyl)-3-dimethylamino-2-propen-1-one (3.1 g), dicyandiamide (1.68 g) and 28% sodium methylate-methanol solution (4 ml) in ethanol (50 ml) was refluxed for 5 hours under stirring.

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The reaction mixture was evaporated in vacuo and the residue was dissolved in a mixture of ethyl acetate and water. The resultant solution was adjusted to pH 7.0 with 10% hydrochloric acid. The separated organic layer was washed with brine and dried over magnesium sulfate. The solvent was evaporated in vacuo and the residue was subjected to column chromatography on silica gel and eluted with a mixture of chloroform and methanol (97:3). The eluted solution was evaporated in vacuo and the residue was recrystallized from a mixture of ethyl acetate and diisopropyl ether to give 4,5-bis(4-methoxyphenyl)-2-cyanoaminopyrimidine (2.0 g).

- The following compounds were obtained according to a similar manner to that of Example 18-(1).
  - (2) 2-Acetylaminomethyl-4,5-bis(4-methoxyphenyl)pyrimidine

25 NMR (DMSO-d<sub>6</sub>, δ): 1.95 (3H, s), 3.64 (3H, s), 3.67 (3H, s), 4.50 (2H, d, J=6Hz), 6.83 (2H, d, J=9Hz), 6.90 (2H, d, J=9Hz), 7.15 (2H, d, J=9Hz), 7.38 (2H, d, J=9Hz), 8.45 (1H, m), 8.60 (1H, s)

(3) 4,5-Bis(4-methoxyphenyl)-2-(1-methyl-1,2,3,6tetrahydropyridin-4-yl)pyrimidine
mp: 131-132.5°C
IR (Nujol): 1650, 1605, 1580, 1560, 1500 cm<sup>-1</sup>

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## Example 19

(1) A mixture of 1,2-bis(4-methoxyphenyl)-3dimethylamino-2-propen-1-one (3.1 g), 4-amidino-morpholine hydrobromide (4.2 g) and 28% sodium methylate-methanol solution (4 ml) in ethanol (30 ml) was refluxed for 7 5 hours under stirring. The reaction mixture was evaporated in vacuo and the residue was dissolved in a mixture of ethyl acetate and 10% hydrochloric acid. separated-aqueous layer was adjusted to pH 8.0 with 20% aqueous potassium carbonate and evaporated with ethyl 10 The extract was washed with brine and dried over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo and the residue was recrystallized from a mixture of ethyl acetate and diisopropyl ether to give 4,5-bis(4-methoxyphenyl)-2-morpholino pyrimidine 15 (0.47 g).

mp: 147-150°C

IR (Nujol): 1605, 1585, 1570, 1525 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 3.73 (6H, s), 3.73 (8H, s), 6.80 (2H, d, J=8Hz), 6.85 (2H, d, J=9Hz), 7.05 (2H, d, J=9Hz), 7.32 (2H, d, J=9Hz), 8.26 (1H, s)

The following compounds were obtained according to a similar manner to that of Example 19-(1).

30 (2) 4,5-Bis(4-methoxyphenyl)-2-(pyridin-4-yl)pyrimidine
mp: 175-177°C
IR (Nujol): 1600, 1590, 1565, 1550, 1500 cm<sup>-1</sup>
NMR (DMSO-d<sub>6</sub>, δ): 3.80 (6H, s), 6.90 (2H, d, J=9Hz),
6.95 (2H, d, J=9Hz), 7.25 (2H, d, J=9Hz),

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7.50 (2H, d, J=9Hz), 8.30 (2H, dd, J=2, 5Hz),
                 8.78 (2H, dd, J=2, 5Hz), 8.83 (1H, s)
            MASS (m/z): 370 (M^++1)
 5
       (3) 4,5-Bis(4-methoxyphenyl)-2-(2,6-dimethylmorpholino)-
            pyrimidine
            mp: 138-140°C
            IR (Nujol): 1610, 1590, 1570, 1500 cm<sup>-1</sup>
            NMR (DMSO-d_6, \delta): 1.17 (6H, d, J=6Hz),
10
                 2.30-2.70 (2H, m), 3.35-3.80 (2H, m), 3.73 (6H,
                 s), 4.40-4.72 (2H, m), 8.78 (2H, d, J=9Hz),
                 6.82 (2H, d, J=9Hz), 7.03 (2H, d, J=9Hz),
                 7.32 (2H, d, J=9Hz), 8.32 (1H, s)
            MASS (m/z): 405 (M^{+})
            Elemental Analysis Calcd. for C_{24}H_{27}N_3O_3:
15
                                         C 71.09, H 6.71, N 10.36
                                Found: C 70.82, H 6.69, N 10.30
```

## Example 20

20 A solution of 4,5-bis(4-methoxyphenyl)-2-(pyridin-4yl)pyrimidine (2.6 g) and 3-fluorobenzyl iodide (2.36 g) in a mixture of chloroform and methanol (2:4)(30 ml) was allowed to stand at ambient temperature for 2 days. reaction mixture was evaporated in vacuo and the residue 25 containing 4-[4,5-bis(4-methoxyphenyl)pyrimidin-2-yl]-1-(3-fluorobenzyl)pyridinium iodide was dissolved in a mixture of methanol (40 ml), chloroform (10 ml) and water (10 ml). To the resulting solution was added portionwise sodium borohydride (0.54 g) with stirring at 5 to 10°C. 30 The reaction mixture was stirred for one hour at the same The reaction mixture was evaporated in temperature. The residue was dissolved in chloroform and washed with brine and dried over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo and the residue was subjected to column chromatography on silica 35

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gel and eluted with a mixture of chloroform and acetone (19:1). The eluted solution was concentrated in vacuo and the crystalline residue was collected by filtration and dried in vacuo to give 4,5-bis(4-methoxyphenyl)-2-[1-(3-fluorobenzyl)-1,2,3,6-tetrahydropyridin-4-yl]pyrimidine (1.33 g).

mp: 130-132°C
IR (Nujol): 1660, 1615, 1560, 1505 cm<sup>-1</sup>
NMR (DMSO-d<sub>6</sub>, δ): 2.55-2.77 (4H, m), 3.03-3.27 (2H, m), 3.60 (2H, s), 3.71 (3H, s), 3.77 (3H, s), 6.77 (2H, d, J=9Hz), 6.85 (2H, d, J=9Hz), 7.10 (2H, d, J=9Hz), 7.31 (2H, d, J=9Hz), 6.70-7.40 (5H, m), 8.53 (1H, s)

#### 15 Example 21

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(1) A solution of 4,5-bis(4-methoxyphenyl)-2-(pyridin-4-yl)pyrimidine (3.7 g) and methyl iodide (4 ml) in a mixture of chloroform and methanol (9:1) (30 ml) was allowed to stand at ambient temperature for 2 days. reaction mixture was evaporated in vacuo and the residue was dissolved in a mixture of methanol (50 ml) and water (10 ml). To the resultant solution was added portionwise sodium borohydride (0.76 g) under stirring at 5 to 10°C. The reaction mixture was stirred for one hour at the same temperature. Water (16 ml) was added to the reaction mixture and the precipitate was collected by filtration. The precipitate was dissolved in chloroform and washed with brine and dried over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo, and the residue was recrystallized from an aqueous methanol to give 4,5-bis(4-methoxyphenyl)-2-(1-methyl-1,2,3,6tetrahydropyridin-4-yl)pyrimidine (1.7 g).

mp : 131-132.5°C IR (Nujol) : 1650, 1605, 1580, 1560, 1500 cm<sup>-1</sup> NMR (DMSO-d<sub>6</sub>,  $\delta$ ) : 2.30 (3H, s), 2.40-2.80 (4H, m),

```
3.00-3.20 (2H, m), 3.75 (3H, s), 3.77 (3H, s), 6.83 (2H, d, J=9Hz), 6.89 (2H, d, J=9Hz), 7.13 (2H, d, J=7Hz), 7.18 (1H, s), 7.37 (2H, d, J=9Hz), 8.58 (1H, s)
```

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- (2) 4,5-Bis(4-methoxyphenyl)-2-(1-ethyl-1,2,3,6-tetrahydropyridin-4-yl)pyrimidine was obtained according to a similar manner to that of Example 21-(1). The product obtained above was dissolved in a solution of hydrochloric acid-ethanol and the resultant solution was evaporated in vacuo. The residue was recrystallized from a mixture of ethanol and ether to give 4,5-bis(4-methoxyphenyl)-2-(1-ethyl-1,2,3,6-tetrahydropyridin-4-yl)pyrimidine hydrochloride.

The following compounds were obtained according to a similar manner to that of Example 21-(1).

(3) 2,3-Bis(4-methoxyphenyl)-6-[1-(4-fluorobenzyl)1,2,3,6-tetrahydropyridin-4-yl]pyridine
mp: 115-117°C

IR (Nujol): 1600, 1580, 1545, 1500 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 2.64 (4H, br s), 3.12 (2H, br s),
3.59 (2H, s), 3.73 (3H, s), 3.74 (3H, s),
6.72-7.07 (5H, m), 7.11-7.42 (8H, m),
7.47 (1H, d, J=8Hz), 7.67 (1H, d, J=8Hz)

MASS (m/z): 481 (M<sup>+</sup>)

```
(4) 4,5-Bis(4-methoxyphenyl)-2-[1-(4-fluorobenzyl)-
           1,2,3,6-tetrahydropyridin-4-yl]pyrimidine
           mp : 118-120°C
           IR (Nujol): 1645, 1600, 1570, 1500 cm<sup>-1</sup>
           NMR (DMSO-d_6, \delta): 2.67 (4H, br s), 3.17 (2H, br s),
 5
                3.61 (2H, s), 3.75 (3H, s), 3.77 (3H, s), 6.87
                 (2H, d, J=9Hz), 6.94 (2H, d, J=9Hz), 7.11-7.20
                 (3H, m), 7.13 (2H, d, J=9Hz), 7.33 (2H, d,
                 J=9Hz), 7.20-7.42 (2H, m), 8.64 (1H, s)
10
           4,5-Bis(4-methoxyphenyl)-2-[1-(2-phenylethyl)-
           1,2,3,6-tetrahydropyridin-4-yl]pyrimidine
           mp: 135-136°C
           IR (Nujol): 1600, 1560, 1500 cm<sup>-1</sup>
           NMR (DMSO-d<sub>6</sub>, \delta): 2.70-2.85 (8H, m), 3.27-3.35 (2H,
15
                 m), 3.76 (6H, s), 6.87 (2H, d, J=9Hz), 6.94 (2H,
                 d, J=9Hz), 7.16-7.33 (6H, m), 7.46 (2H, d,
                 J=9Hz), 8.65 (1H, s)
            MASS (m/z): 477 (M^+)
20
           4,5-Bis(4-methoxyphenyl)-2-[1-(2-fluorobenzyl)-
       (6)
            1,2,3,6-tetrahydropyridin-4-yl]pyrimidine
            mp: 96-99°C
            IR (Nujol): 1650, 1600, 1570, 1550 cm<sup>-1</sup>
            NMR (DMSO-d<sub>6</sub>, \delta): 2.69 (4H, br s), 3.23 (2H, br s),
 25
                 3.68 (2H, s), 3.75 (3H, s), 3.77 (3H, s), 6.86
                  (2H, d, J=9Hz), 6.93 (2H, d, J=9Hz), 7.18 (2H,
                  d, J=9Hz), 7.13-7.47 (5H, m), 7.39 (2H, d,
                  J=9Hz), 8.67 (1H, s)
            MASS (m/z) : 481 (M^+)
 30
       (7) 4,5-Bis(4-methoxyphenyl)-2-[1-benzyl-1,2,3,6-
            tetrahydropyridin-4-yl]pyrimidine
```

IR (Nujol): 1645, 1600, 1570, 1545 cm<sup>-1</sup>

mp: 128-130°C

NMR (DMSO-d<sub>6</sub>, δ): 2.50 (4H, br s), 3.20 (2H, br s), 3.63 (2H, s), 3.75 (3H, s), 3.77 (3H, s), 6.87 (2H, d, J=9Hz), 6.93 (2H, d, J=9Hz), 7.17 (2H, d, J=9Hz), 7.16-7.42 (5H, m), 7.25 (1H, s), 7.39 (2H, d, J=9Hz), 8.64 (1H, s)

MASS (m/z): 463 (M<sup>+</sup>)

#### Example 22

A mixture of 4,5-bis(4-methoxyphenyl)-(1-methyl-1,2,3,6-tetrahydropyridin-4-yl)pyrimidine (2.10 g), 10% 10 palladium-carbon (0.6 g) and ammonium formate (1.71 g) in acetic acid (40 ml) was stirred at 100 to 110°C for 2 hours. The reaction mixture was filtered and the filtrate was evaporated in vacuo. The residue was dissolved in a 15 mixture of ethyl acetate, tetrahydrofuran and water and the resulting mixture was adjusted to pH 8.0 with 20% aqueous potassium carbonate. The separated organic layer was washed with brine and dried over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo and 20 the residue was subjected to column chromatography on silica gel and eluted with a mixture of chloroform and methanol (8:2). The eluted solution was evaporated in vacuo and the oily residue was dissolved in a solution of hydrochloric acid-ethanol and the resultant solution was 25 evaporated in vacuo and the residue was recrystallized from a mixture of ethanol and ether to give 4,5-bis(4methoxyphenyl)-2-(1-methylpiperidin-4-yl)pyrimidine dihydrochloride (0.34 g).

mp: 227-230°C (dec.)

IR (Nujol): 1590, 1510 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 2.02-2.20 (4H, m), 2.72 (3H, s),

2.96-3.64 (6H, m), 3.77 (5H, s), 6.90 (2H, d,

J=9Hz), 6.96 (2H, d, J=9Hz), 7.21 (2H, d,

J=9Hz), 7.42 (2H, d, J=9Hz), 8.72 (1H, s)

The following compound was obtained according to a similar manner to that of Example 22-(1).

```
4,5-Bis(4-methoxyphenyl)-2-[1-(3-fluorobenzyl)-

piperidin-4-yl]pyrimidine hydrochloride

mp: 207-211°C (dec.)

IR (Nujol): 1595, 1510 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 2.16-2.22 (4H, m), 3.00-3.54 (5H, m), 3.77 (2H, s), 4.39 (2H, d, J=5Hz),

6.85 (2H, d, J=9Hz), 6.96 (2H, d, J=9Hz),

7.20 (2H, d, J=9Hz), 7.41 (2H, d, J=9Hz),

7.27-7.75 (4H, m), 8.70 (1H, s)
```

#### Example 23

A solution of 4,5-bis(4-methoxyphenyl)-2-15 cyanoaminopyrimidine (1.0 g) and morpholine (1.3 g) in ethanol (10 ml) was refluxed for one hour under stirring. The reaction mixture was evaporated in vacuo and the residue was dissolved in a mixture of ethyl acetate and 10% hydrochloric acid. The separated aqueous layer was 20 adjusted to pH 8.0 with 20% aqueous potassium carbonate and extracted with a mixture of ethyl acetate and tetrahydrofuran. The organic layer was washed with brine and dried over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo and the residue was 25 recrystallized from a mixture of ethyl acetate and ether to give 4,5-bis(4-methoxyphenyl)-2-[(4morpholinecarboximidoyl)amino]pyrimidine (0.72 g).

```
mp: 176-178°C

IR (Nujol): 3270. 3150, 1610, 1590 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 3.33 (4H, s), 3.60 (4H, s),

3.75 (6H, s), 6.83 (2H, d, J=9Hz),

6.85 (2H, d, J=9Hz), 7.08 (2H, d, J=7Hz),

7.28 (2H, d, J=9Hz), 8.36 (1H, s), 8.37 (2H, s)

MASS (m/z): 419 (M<sup>+</sup>)
```

The following compounds were obtained according to a similar manner to that of example 23-(1).

- (4) 4,5-Bis(4-methoxyphenyl)-2-[3-(1-benzylpiperidin-4-yl)guanidino]pyrimidine
   mp: 157-159°C (dec.)
   IR (Nujol): 3150, 3300, 1640 (sch), 1605, 1580 cm<sup>-1</sup>
   NMR (DMSO-d<sub>6</sub>, δ): 1.17-1.52 (2H, m), 1.81-2.28 (4H, m), 2.67-2.73 (2H, m), 3.44 (2H, s), 3.75 (6H, s), 3.64-3.89 (1H, m), 6.82 (2H, d, J=9Hz),

20

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6.91 (2H, d, J=9Hz), 7.11 (2H, d, J=9Hz), 7.19-7.40 (8H, m), 8.48 (1H, s)

MASS (m/z): 522 (M<sup>+</sup>)
```

5 (5) 4,5-Bis(4-methoxyphenyl)-2-[(4-benzyl-1-piperidine-carboximidoyl)amino]pyrimidine

mp: 160°C (dec.)

IR (Nujol): 3300, 3150, 1600, 1580, 1520 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 1.02-1.15 (2H, m), 1.57-1.78 (3H, m), 2.70-2.82 (2H, m), 3.22-3.86 (2H, m), 3.75 (6H, s), 4.33-4.40 (2H, m), 6.83-6.90 (4H, m), 7.07 (2H, d, J=9Hz), 7.10-7.31 (8H, m), 8.33 (1H, s), 8.37 (2H, br s)

MASS (m/z) : 507  $(M^+)$ 

(6) 4,5-Bis(4-methoxyphenyl)-2-[{4-(2-methoxyphenyl)-1piperazinecarboximidoyl}amino]pyrimidine

mp: 199-200°C

IR (Nujol): 3320, 3180, 1605, 1570, 1515 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>,δ): 3.00 (4H, br s), 3.75 (9H, s),

3.82(4H, br s), 6.84-7.07 (8H, m), 7.15 (2H, d, J=9Hz), 7.31 (2H, d, J=9Hz), 8.38 (1H, s), 8.49 (2H, br s)

# 25 <u>Example 24</u>

- (1) A solution of 4,5-bis(4-methoxyphenyl)-2-cyanoamino pyrimidine (1.0 g) and piperidine (1.1 g) in methyl cellosolve (5 ml) was stirred at 120 to 125°C for 3 hours. The reaction mixture was poured into water and extracted
- with ethyl acetate. The extract was washed with brine and dried over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo and the residue was recrystallized from a mixture of ethyl acetate and disopropyl ether to give 4,5-bis(4-methoxyphenyl)
  2-[(1-piperidinecarboximidoyl)amino]pyrimidine (0.65 g).

```
mp: 185-187°C
           IR (Nujol): 3330, 3170, 1610, 1590, 1565, 1525 cm<sup>-1</sup>
           NMR (DMSO-d_6, \delta): 1.30-1.70 (6H, m), 3.40-3.70 (4H,
                m), 3.73 (6H, s), 6.81 (2H, d, J=9Hz), 6.83 (2H,
                d, J=9Hz), 7.05 (2H, d, J=9Hz), 7.27 (2H, d,
5
                 J=9Hz), 8.30 (1H, s), 8.32 (2H, m)
           The following compounds were obtained according to a
      similar manner to that of Example 24-(1).
10
            4,5-Bis(4-methoxyphenyl)-2-(3-cyclohexylguanidino)-
       (2)
            pyrimidine
            mp: 203-205°C
            IR (Nujol): 3310, 3205, 1595, 1510 cm^{-1}
            NMR (DMSO-d_6, \delta): 0.86-2.03 (11H, m), 3.75 (6H, s),
15
                 6.85 (2H, d, J=9Hz), 6.88 (2H, d, J=9Hz),
                 7.10 (2H, d, J=9Hz), 7.30 (2H, d, J=9Hz),
                 8.47 (lH, s)
            MASS (m/z): 431 (M^+)
20
            4,5-Bis(4-methoxyphenyl)-2-[(2-imidazolin-2-yl)-
       (3)
            amino]pyrimidine
            mp: 190-192°C
            IR (Nujol): 3460, 3320, 3200, 1640, 1610, 1585,
                           1575, 15105 cm<sup>-1</sup>
25
            NMR (DMSO-d_6, \delta): 3.10-3.45 (4H, m), 3.73 (6H, s),
                  6.60 (2H, s), 6.80 (2H, d, J=9Hz), 6.83 (2H, d,
                 J=9Hz), 7.03 (2H, d, J=9Hz), 7.28 (2H, d,
                  J=9Hz), 8.15 (1H, s)
30
       (4) 4,5-Bis(4-methoxyphenyl)-2-(3-isopropylguanidino)-
            pyrimidine
            mp : 167-169°C (dec.)
             IR (Nujol): 3350, 3170, 1680, 1640, 1610, 1580,
                           1530 cm<sup>-1</sup>
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NMR (DMSO-d_6, \delta): 1.18 (6H, d, J=5Hz), 3.40-4.00
          (lH, m), 3.75 (6H, s), 6.60-7.50 (8H, m), 8.46
          (lH, s)
    MASS (m/z) : 391 (M^+)
(5) 4,5-Bis(4-methoxyphenyl)-2-(3-benzylguanidino)-
```

pyrimidine

mp : 204-206°C IR (Nujol): 3460, 1630, 1610, 1560 cm<sup>-1</sup> NMR (DMSO- $d_6$ ,  $\delta$ ): 3.80 (6H, s), 4.62 (2H, s), 10 6.92 (2H, d, J=9Hz), 6.97 (2H, d, J=9Hz), 7.20 (2H, d, J=9Hz), 7.38 (2H, d, J=9Hz), 7.43 (5H, s), 8.43 (1H, s)

#### Example 25 15

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A solution of 4,5-bis(4-methoxyphenyl)-2-cyanoamino pyrimidine (0.67 g) and dimethylamine hydrochloride (0.4 g) in ethanol (20 ml) was refluxed for 10 hours under stirring. The reaction mixture was evaporated in vacuo and the residue was dissolved in a mixture of ethyl acetate and 10% hydrochloric acid. The separated aqueous layer was adjusted to pH 8.0 with 20% aqueous potassium carbonate and extracted with ethyl acetate. The extract was washed with brine and dried over magnesium sulfate. After filtration, the filtrate was evaporated in vacuo and the oily residue was dissolved in a hydrochloric acid ethanol solution. The resultant mixture was evaporated in vacuo and the residue was recrystallized from a mixture of ethanol and ether to give 4,5-bis(4-methoxyphenyl)-2-(3,3-

dimethylguanidino)pyrimidine hydrochloride (0.28 g). mp: 238-239°C IR (Nujol): 3130, 1640, 1605, 1580 cm<sup>-1</sup> NMR (DMSO- $d_6$ ,  $\delta$ ): 3.20 (6H, s), 3.78 (6H, s), 6.88 (2H, d, J=7Hz), 6.95 (2H, d, J=9Hz), 7.17 (2H, d, J=9Hz), 7.39 (2H, d, J=9Hz),

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8.62 (lH, s), 9.22 (2H, m)

MASS (m/z): 377 (M<sup>+</sup>)

Elemental Analysis Calcd. for C<sub>21</sub>H<sub>23</sub>N<sub>5</sub>O<sub>2</sub>:

C 60.94, H 5.84, N 16.92, C1 8.57

Found: C 60.90, H 5.64, N 16.78, C1 8.64

### Example 26

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A mixture of 2,3-bis(4-methoxyphenyl)pyrazine-5-carboxylic acid (1.01 g), 1-hydroxybenzotriazole (446 mg) and 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (573 mg) in N,N-dimethylformamide (20 ml) was stirred for 1 hour at ambient temperature. N-methyl piperazine (500 mg) was added thereto, and the mixture was stirred for 2 hours. After the solvent was removed, to the residue were added chloroform (30 ml) and water (20 ml), and the mixture was adjusted to pH 9 with a saturated aqueous solution of potassium carbonate. The separated organic layer was washed with water (20 ml), dried over magnesium sulfate and concentrated. The residue was dissolved with ethyl acetate (20 ml) and crystallized with a diethyl ether solution saturated with hydrogen chloride and the precipitate was collected and washed with ethyl acetate to give 2,3-bis(4-methoxyphenyl)-5-[(4methylpiperazin-l-yl)carbonyl}pyrazine hydrochloride.

25 mp: 90-93°C

IR (Nujol): 1630, 1600, 1505 cm<sup>-1</sup>

NMR (D<sub>2</sub>O, δ): 3.23 (3H, s), 3.70 (6H, s),

3.1-4.3 (8H, m), 6.6-7.1 (4H, m), 7.2-7.7 (4H,

m), 9.03 (1H, s)

MASS (m/z): 418 (M<sup>+</sup> of free compound)

#### Example 27

(1) A mixture of 2,3-bis(4-methoxyphenyl)pyrazine-5-carboxylic acid (1.03 g), l-hydroxybenzotriazole (446 mg) and l-ethyl-3-(3-dimethylaminopropyl)carbodiimide

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hydrochloride (573 mg) in N,N-dimethylformamide (20 ml) was stirred for 1 hour at ambient temperature. 30% Methylamine-methanol solution (5 ml) was added thereto, and the mixture was stirred for 3 hours. After the solvent was removed, to the residue were added chloroform (30 ml) and water (20 ml), and the mixture was adjusted to pH 9 with a saturated aqueous solution of potassium carbonate. The separated organic layer was dried over magnesium sulfate and concentrated. residue was subjected to column chromatography on silica gel (50 g) and eluted with 15% solution of methanol in chloroform. The fractions containing the desired compound were collected and concentrated. The residue was crystallized with a diethyl ether solution saturated with hydrogen-chloride to give 2,3-bis(4-methoxyphenyl)-5-15 (methylcarbamoyl)pyrazine hydrochloride.

mp: 125-127°C IR (Nujol): 1680, 1600, 1520, 1510 cm<sup>-1</sup> NMR (DMSO- $d_6$ ,  $\delta$ ): 2.87 (3H, d, J=5Hz), 3.77 (6H, s), 6.93 (4H, d, J=10Hz), 7.3-7.6 (4H, m), 8.72 (lH, d, J=5Hz), 9.03 (lH, s) MASS (m/z): 349  $(M^+)$  of free compound)

The following compound was obtained according to a similar manner to that of Example 27-(1). 25

2,3-Bis(4-methoxyphenyl)-5-[(1-benzylpiperidin-4-yl)carbamoyl]pyrazine mp: 157-159°C IR (Nujol): 3400, 1655, 1600, 1500 cm<sup>-1</sup> 30 NMR (CDCl<sub>3</sub>,  $\delta$ ): 1.5-2.33 (6H, m), 2.77-3.0 (2H, m), 3.53 (2H, s), 3.82 (3H, s), 3.85 (3H, s), 3.95-4.2 (1H, m), 6.8-7.02 (4H, m), 7.15-7.53 (10H, m), 7.76 (1H, d, J=8.5Hz), 9.28 (1H, s)

MASS (m/z) : 508  $(M^+)$ 35

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## Example 28

A mixture of 5-chloro-2,3-(4-methoxyphenyl)pyrazine (653 mg) and guanidine carbonate (900 mg) in N,N-dimethyl formamide (5 ml) was heated at 140°C for 15 hours. After the solvent was removed under reduced pressure, to the residue were added water (15 ml) and 5% methanol in chloroform (30 ml). The separated organic layer was washed with saturated aqueous solution of sodium chloride and dried over magnesium sulfate. The solvent was removed and the residue was subjected to column chromatography on silica gel (50 g) and eluted with 10% solution of methanol in chloroform, the fractions containing the desired compound were collected and concentrated. The residue was pulverized with diethyl ether to give

5-(N,N-dimethylamino)-2,3-bis(4-methoxyphenyl)pyrazine.

mp: 161-163°C

IR (Nujol): 1605, 1555, 1490 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 3.10 (6H, s), 3.68 (6H, s),

6.80 (2H, d, J=10Hz), 6.82 (2H, d, J=10Hz),

7.20 (2H, d, J=10Hz), 7.30 (2H, d, J=10Hz),

8.07 (1H, s)

MASS (m/z): 335 (M<sup>+</sup>)

## Example 29

25 A mixture of 5-chloro-2,3-bis(4-methoxyphenyl)pyrazine (980 mg) and guanidine (1.8 g) and
N,N-dimethylformamide (1.0 ml) was heated at 150°C for 8
hours. After the solvent was removed under reduced
pressure, the residue was subjected to column
30 chromatography on silica gel (50 g) and eluted with 15%
solution of methanol in chloroform. The fractions
containing the desired compound were collected and
concentrated. The residue was dissolved in ethyl acetate
(20 ml) and crystallized with an ethanol solution saturated
with hydrogen chloride. The precipitate was collected and

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washed with ethyl acetate to give 5-guanidino-2,3-bis(4methoxyphenyl)pyrazine hydrochloride.

mp : 246-269°C (dec.) IR (Nujol): 3250, 1680, 1620, 1600, 1510 cm<sup>-1</sup> NMR (DMSO- $d_6$ ,  $\delta$ ): 3.74 (6H, s), 6.7-7.0 (4H, m), 7.2-7.5 (4H, m), 8.38 (5H, s) 5 349 (M<sup>+</sup>) MASS (m/z):

## Example 30

A mixture of morpholine (3 ml) and 5-chloro-2,3-bis(4-methoxyphenyl)pyrazine (980 mg) was 10 stirred at 100°C for 3 hours. The solvent was removed under reduced pressure and the residue was crystallized with ether to give 2,3-bis(4-methoxyphenyl)-5-

(morpholin-4-yl)-pyrazine. 15

mp: 138-139°C IR (Nujol): 1603, 1573, 1545, 1505 cm<sup>-1</sup> NMR (CDCl<sub>3</sub>,  $\delta$ ): 3.5-4.0 (8H, m), 3.76 (6H, s), 6.89 (4H, d, J=10Hz), 7.2-7.5 (4H, m), 8.03 (lH, s) 20 MASS (m/z): 377  $(M^{+})$ 

## Example 31

A mixture of piperazine (5 g) and 5-chloro-2,3-bis(4-methoxyphenyl)pyrazine (980 mg) in ethanol (10 ml) was refluxed for 3 hours. After the solvent was 25 removed, to the residue were added ethyl acetate (20 ml) and water (30 ml). The separated organic layer was washed with a saturated aqueous solution of sodium chloride and dried over magnesium sulfate. After the solvent was removed, the residue was crystallized with ether to give 30 2,3-bis(4-methoxyphenyl)-5-(piperazin-l-yl)pyrazine.

128-129°C IR (Nujol): 3400, 1605, 1570, 1545, 1510 cm<sup>-1</sup> NMR (DMSO- $d_6$ ,  $\delta$ ): 2.6-2.9 (4H, m), 3.4-3.6 (4H, m), 35

3.73 (6H, s), 6.82 (4H, d, J=9Hz), 7.12-7.43 (4H, m), 8.12 (1H, s)

MASS (m/z): 376 (M<sup>+</sup>)

#### 5 Example 32

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A mixture of 5-chloro-2,3-bis(4-methoxyphenyl)pyrazine (980 mg) and N-methylpiperazine (3 ml) was stirred at 60°C for 3 hours and at 100°C for 2 hours. After the solvent was removed under reduced pressure, the residue was dissolved in diethyl ether (30 ml) and crystallized with an ether solution saturated with hydrogen chloride. The precipitate was collected and washed with ether to give 2,3-bis(4-methoxyphenyl)-5-(4-methylpiperazin-l-yl)-pirazine dihydrochloride.

Example 33

A mixture of 2,3-bis(4-methoxyphenyl)-5-methyl pyrazine (1.07 g), dimethylamine hydrochloride (346 mg) and paraformaldehyde (126 mg) in acetic acid (3 ml) was stirred for 10 days at ambient temperature. After the solvent was removed, to the residue were added water (10 ml) and 5% solution of methanol in chloroform (20 ml), and the mixture was adjusted to pH 9 with a saturated aqueous solution of potassium carbonate. The separated organic layer was dried over magnesium sulfate. The solvent was removed and the residue was subjected to column chromatography on silica gel (50 g) and eluted with 10% solution of methanol in chloroform. The fractions containing the desired compound were collected and concentrated. The residue was dissolved in ethyl acetate

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(30 ml) and crystallized with a solution of hydrogen
     chloride in ether to give 5-[2-(N,N-dimethylamino)ethyl]-
      2,3-bis(4-methoxyphenyl)pyrazine hydrochloride.
           mp : 203-207°C (dec.)
           IR (Nujol): 3350, 2600, 1600, 1520 cm<sup>-1</sup>
5
           NMR (D_2O, \delta): 2.8-3.2 (2H, m), 3.0 (6H, s),
                3.6-4.0 (2H, m), 3.85 (6H, s), 6.8-7.6 (8H, m),
                8.83 (lH, s)
           MASS (m/z): 363 (M^+)
10
      Example 34
           A mixture of 3-(2-t-butoxycarbonylaminoethyl)-5,6-
      bis(4-methoxyphenyl)-1,2,4-triazine (5.00 g) and
      2.5-norbornadiene (12.0 ml) in xylene (200 ml) was
      refluxed under stirring for 3 days. The reaction mixture
15
      was evaporated in vacuo. To the residue was added
       4N-1,4-dioxane solution of hydrogen chloride and the
       mixture was stirred at 0°C for 2 hours. The reaction
       mixture was evaporated in vacuo. The resulting
       precipitate was washed with diethyl ether and dried to
20
       give 6-(2-aminoethyl)-2,3-bis(4-methoxyphenyl)pyridine
       hydrochloride (5.35 g).
            IR (Nujol): 1610, 1510, 1300, 1250 cm<sup>-1</sup>
            NMR (DMSO-d_6, \delta): 3.33-3.43 (4H, m), 3.75 (3H, s),
                  3.78 (3H, s), 6.91 (2H, d, J=7Hz), 6.94 (2H, d,
 25
                  J=7Hz), 7.11 (2H, d, J=7Hz), 7.34 (2H, d,
                  J=7Hz), 7.76 (lH, d, J=8Hz), 8.19 (lH, d,
                  J=8Hz), 8.44 (3H, br s)
             MASS (m/z): 334 (M^+) of free compound)
 30
        E mple 35
        (1) A mixture of 3-{N,N-dimethylaminomethyl}-5,6-bis(4-
        methoxyphenyl)-1,2,4-triazine (0.60 g) and an ethanol
        solution of hydrogen chloride (15 ml) was stirred at the
```

ambient temperature for 3 hours. The reaction mixture was

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evaporated in vacuo and the residue was washed with diethyl ether to give 3-{N,N-dimethylaminomethyl}-5,6-bis(4-methoxyphenyl)-1,2,4-triazine hydrochloride (0.58g).

5 mp: 215-218°C
IR (Nujol): 1600, 1300, 1250 cm<sup>-1</sup>
NMR (DMSO-d<sub>6</sub>, δ): 2.97 (3H, s), 2.99 (3H, s), 3.80
(3H, s), 3.81 (3H, s), 4.85 (2H, d, J=6Hz), 6.98
(2H, d, J=8Hz), 7.03 (2H, d, J=8Hz), 7.50 (2H,
d, J=8Hz), 7.63 (2H, d, J=8Hz)

Mass (m/z): 350 (M<sup>+</sup> of free compound)

The following compound was obtained according to a similar manner to that of Example 35-(1)

(2) 2,3-Bis(4-methoxyphenyl)-6-(N,N-dimethylaminomethyl)pyridine dihydrochloride

mp: 80°C-82°C (dec.)
IR (Nujol): 1600, 1510, 1300, 1250 cm<sup>-1</sup>

NMR (DMSO-d<sub>6</sub>, δ): 2.86 (6H, s), 3.75 (3H, s), 3.76
(3H, s), 4.54 (2H, s), 6.85 (2H, d, J=8Hz), 6.91
(2H, d, J=8Hz), 7.13 (2H, d, J=8Hz), 7.34 (2H, d, J=8Hz), 7.72 (1H, d, J=8Hz), 7.90 (1H, d,

J=8Hz)

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## CLAIMS

1. A compound of the formula:

wherein R<sup>1</sup> and R<sup>2</sup> are each lower alkoxy,
R<sup>3</sup> is heterocyclic group selected from the
group consisting of pyridyl,
tetrahydropyridyl, piperidyl,
piperazinyl and morpholinyl, which may
have suitable substituent(s);
substituted amino;
carboxy(lower)alkenyl;
carboxy(lower)alkyl;
hydroxy(lower)alkyl;
amino(lower)alkyl which may have
suitable substituent(s);

a group of the formula :

(in which R<sup>4</sup> is hydrogen, ethoxy,
mono(or di)lower
alkylamino, di(lower)alkylamino-(lower)alkylamino,
heterocyclicamino
which may have suitable
substituent(s), or

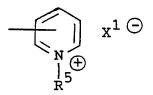
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heterocyclic group which
may have suitable
substituent(s));

or a group of the formula :

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Y is CH or N and Z is CH or N,

with proviso that

when R<sup>3</sup> is pyridyl; piperidyl which may have hydroxy
group; piperazinyl which has lower alkyl
group or hydroxy(lower)alkyl group;
morpholinyl; lower alkenylamino;
hydroxy(lower)alkylamino; phenylamino which
may have lower alkoxy group or halogen on
the benzene ring; halophenyl(lower)alkylamino; phenylsulfonylamino which has
nitro group, amino group or halogen on the
benzene ring; or
amino substituted with two substituents

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selected from the group consisting of lower alkyl and hydroxy(lower)alkyl, and

Y is N,

then Z is CH,

and a pharmaceutically acceptable salt thereof.

A compound of claim 1, wherein 2. R<sup>3</sup> is heterocyclic group selected from the group consisting of pyridyl, tetrahydropyridyl, piperidyl, piperazinyl and morpholinyl, which may have one to three suitable substituent(s); 5 cyanoamino; imidazolinylamino; guanidino; di(lower)alkylguanidino; lower alkylguanidino; cyclo(lower)alkylguanidino; ar(lower)alkylguanidino; heterocyclicguanidino which may be substituted with one to three 10 suitable substituent(s); (1-heterocyclic-1-iminomethyl)amino which may have one to three suitable substituent(s); di(lower)alkylamino; hydroxy(lower)alkyl; carboxy(lower)alkyl; carboxy(lower)alkenyl; 15 amino(lower)alkyl which may have one to three substituent(s) selected from the group consisting of lower alkyl and acyl; a group of the formula : 20

O |<del>|</del>

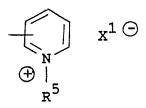
(in which R<sup>4</sup> is hydrogen; ethoxy; mono(or
di)lower alkylamino; di(lower)alkylamino-(lower)alkylamino;
heterocyclicamino which may have
one to three substituent(s)
selected from the group consisting
of lower alkyl, lower alkoxy,
halogen and ar(lower)alkyl; or
heterocyclic group which may have
one to three substituent(s)
selected from the group consisting
of lower alkyl, lower alkoxy,

halogen, ar(lower)alkyl and
hydroxy(lower)alkyl);

or a group of the formula :

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(in which R<sup>5</sup> is lower alkyl, or ar(lower)alkyl which may have one to three substituent(s) selected from the group consisting of halogen, lower alkyl and lower alkoxy; and X<sup>1</sup> is an acid residue).

A compound of claim 2, wherein 3.  $\mathbb{R}^3$  is heterocyclic group selected from the group consisting of pyridyl, tetrahydropyridyl, piperidyl, piperazinyl and morpholinyl, which 20 may have one or two substituent(s) selected from the group consisting of lower alkyl and ar(lower)alkyl which may have one or two halogen; cyanoamino; imidazolinylamino; guanidino; di(lower)alkylguanidino; lower 25 alkylguanidino; cyclo(lower)alkylguanidino; phenyl(lower)alkylguanidino; heterocyclicguanidino which may have ar(lower)alkyl; (l-heterocyclic-l-iminomethyl)amino which may have substituent selected from 30 the group consisting of lower alkyl, ar(lower)alkyl and aryl which may have lower alkoxy; di(lower)alkylamino; hydroxy(lower)alkyl; carboxy(lower)alkyl;

carboxy(lower)alkenyl; amino(lower)alkyl which

may have one or two substituent(s) selected from the group consisting of lower alkyl, lower alkanoyl, lower alkylcarbamoyl, lower alkoxycarbonyl and heterocycliccarbonyl which may have one to three suitable substituent(s); a group of the formula:

> 0 || -C-R<sup>4</sup>

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(in which R4 is hydrogen; ethoxy; mono(or di)lower alkylamino; di(lower)alkylamino-(lower)alkylamino; saturated 5 or 15 6-membered heteromonocyclicamino in which heteromonocyclic group contains 1 to 3 nitrogen atom(s), which may have one or two substituent(s) selected from 20 the group consisting of lower alkyl, lower alkoxy, halogen and phenyl(lower)alkyl; saturated 5 or 6-membered heteromonocyclic group 25 containing 1 to 2 oxygen atom(s) and 1 to 3 nitrogen atom(s) which may have one or two substituent(s) selected from the group consisting of lower alkyl, 30 lower alkoxy, halogen, phenyl(lower)alkyl and hydroxy(lower)alkyl; or saturated 5 or 6-membered heteromonocyclic group 35 containing 1 to 3 nitrogen

atom(s), which may have one or
two substituent(s) selected from
the group consisting of lower
alkyl, lower alkoxy, halogen,
phenyl(lower)alkyl and
hydroxy(lower)alkyl);

or a group of the formula :

⊕ | x<sup>1</sup> ∈

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(in which R<sup>5</sup> is lower alkyl, or
phenyl(lower)alkyl which may
have one or two substituent(s)
selected from the group
consisting of halogen, lower
alkyl and lower alkoxy; and
x<sup>1</sup> is halogen).

4. A compound of claim 3, wherein  $\mathbb{R}^3$  is heterocyclic group selected from the group consisting of pyridyl, tetrahydropyridyl. 25 piperidyl, piperazinyl and morpholinyl, which may have one or two substituent(s) selected from the group consisting of lower alkyl, phenyl(lower)alkyl and halophenyl(lower)alkyl; cyanoamino; imidazolinylamino; guanidino; 30 di(lower)alkylguanidino; lower alkylguanidino; cyclo(lower)alkylguanidino; phenyl(lower)alkylquanidino: phenyl(lower)alkylpiperidylguanidino; {morpholinyl(imino)methyl}amino; 35 {piperidyl(imino)methyl}amino which may have

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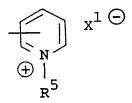
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phenyl(lower)alkyl;
{piperazinyl(imino)methyl}amino which may have
substituent selected from the group consisting
of lower alkyl, phenyl(lower)alkyl and lower
alkoxy-phenyl; di(lower)alkylamino;
hydroxy(lower)alkyl; carboxy(lower)alkyl;
carboxy(lower)alkenyl; amino(lower)alkyl which
may have one or two substituent(s) selected from
the group consisting of lower alkyl, lower
alkanoyl, lower alkylcarbamoyl, lower
alkoxycarbonyl and tetrahydropyridazinylcarbonyl
which may have oxo group;
a group of the formula:

O || |-C-1

or a group of the formula:



5. A compound of claim 4, wherein  $R^3$  is a group of the formula:

5 O | | | -C-R

(in which R<sup>4</sup> is lower alkylpiperazinyl),

Y is N and

Z is N.

10

- 6. A compound of claim 5, which is 5,6-Bis(4-methoxy-phenyl)-3-{(4-methylpiperazin-l-yl)carbonyl}-1,2,4-triazine hydrochloride.
- 7. A compound of claim 4, wherein

  R<sup>3</sup> is lower alkyltetrahydropyridyl,

  Y is CH and
  Z is N.
- 8. A compound of claim 7, which is 4,5-Bis(4-methoxyphenyl)-2-(1-methyl-1,2,3,6tetrahydropyridine-4-yl)pyrimidine.
- 9. A process for preparing a compound of the formula:

$$R^{1}$$
 $Y Z$ 
 $N$ 
 $R^{3}$ 

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wherein R<sup>1</sup> and R<sup>2</sup> are each lower alkoxy,

R<sup>3</sup> is heterocyclic group selected from the group consisting of pyridyl,

tetrahydropyridyl, piperidyl,

piperazinyl and morpholinyl, which may have suitable substituent(s); substituted amino; carboxy(lower)alkenyl; carboxy(lower)alkyl; 5 hydroxy(lower)alkyl; amino(lower)alkyl which may have suitable substituent(s); a group of the formula : 10 | |-C-R<sup>4</sup> (in which R4 is hydrogen, ethoxy, 15 mono(or di)loweralkylamino, di(lower)alkylamino-(lower)alkylamino, heterocyclicamino which may have suitable 20 substituent(s), or heterocyclic group which may have suitable substituent(s)); 25 or a group of the formula : 30 (in which  $R^5$  is lower alkyl or

(in which R° is lower alkyl or ar(lower)alkyl which may have suitable substituent(s), and

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 $X^{1}$  is an acid residue),

Y is CH or N and

Z is CH or N,

with proviso that

when R<sup>3</sup> is pyridyl; piperidyl which may have hydroxy group; piperazinyl which has lower alkyl group or hydroxy(lower)alkyl group; morpholinyl; lower alkenylamino; hydroxy(lower)alkylamino; phenylamino which

may have lower alkoxy group or halogen on the benzene ring; halophenyl(lower)alkylamino; phenylsulfonylamino which has

nitro group, amino group or halogen on the

benzene ring; or

amino substituted with two substituents selected from the group consisting of lower alkyl and hydroxy(lower)alkyl, and

Y is N,

then Z is CH,

or a salt thereof, which comprises

(1) reacting a compound of the formula:

 $\begin{array}{c}
\mathbb{R}^{1} \\
\mathbb{C} = 0 \\
\mathbb{R}^{2}
\end{array}$ 

wherein  $R^1$  and  $R^2$  are each as defined above, with a compound of the formula:

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wherein R<sup>3</sup> is as defined above, or a salt thereof to give a compound of the formula:

wherein  $R^1$ ,  $R^2$  and  $R^3$  are each as defined above, or a salt thereof, or

(2) reacting a compound of the formula:

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above, or its reactive derivative at the carboxy group or a salt thereof with a compound of the formula:

di(lower)alkylamino-(lower)alkylamino,
heterocyclicamino which may have
suitable substituent(s), or

heterocyclic group containing at least one nitrogen atom which may have suitable substituent(s),

or a salt thereof to give a compound of the formula :

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wherein  $R^1$ ,  $R^2$ , Y, Z and -Nare each as defined

or a salt thereof, or

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(3) subjecting a compound of the formula:

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above and  $R^3$  is protected amino(lower)alkyl, 25 or a salt thereof to elimination reaction of the amino-protective group to give a compound of the formula:

30

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above and  $R_b^3 \text{ is amino(lower)alkyl,}$ 

or a salt thereof,

5 or

(4) subjecting a compound of the formula:

wherein  $R^1$ ,  $R^2$ ,  $R_b^3$ , Y and Z are each as defined above.

or its reactive derivative at the amino group

or a salt thereof to acylation reaction to give
a compound of the formula:

$$R^{1}$$
 $Y Z$ 
 $N$ 
 $R^{3}$ 

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above and  $R_C^3$  is acylamino(lower)alkyl,

or a salt thereof,

or

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25

(5) reacting a compound of the formula:

$$\begin{array}{c}
\mathbb{R}^1 \\
\mathbb{Y}^{\frac{1}{2}1} \\
\mathbb{N}
\end{array}$$

10

5

wherein  $R^1$  and  $R^2$  are each as defined above,  $Y^1$  is CH or N and  $Z^1$  is CH or N,

or a salt thereof with a compound of the formula :

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$$x^1-R^5$$

wherein  $R^5$  and  $x^1$  are each as defined above, to give a compound of the formula :

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wherein  $R^1$ ,  $R^2$ ,  $R^5$ ,  $Y^1$ ,  $Z^1$  and  $X^1$  are each as defined above,

or a salt thereof,

or

(6) subjecting a compound of the formula:

25

wherein  $R^1$ ,  $R^2$ ,  $R^5$ , Y, Z and  $X^1$  are each as defined above, or a salt thereof to reduction reaction to give a compound of the formula :

wherein  $R^1$ ,  $R^2$ ,  $R^5$ , Y and Z are each as defined above and

==== means a single or double bond, or a salt thereof, or

(7) subjecting a compound of the formula:

wherein  $R^1$ ,  $R^2$ ,  $R^5$ , Y and Z are each as defined above,

or a salt thereof to reduction reaction to give a compound of the formula :

5

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wherein  $R^1$ ,  $R^2$ ,  $R^5$ , Y and Z are each as defined above, or a salt thereof, or

(8) reacting a compound of the formula:

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carboxy(lower)alkyl;

hydroxy(lower)alkyl; amino(lower)alkyl which may have suitable substituent(s); a group of the formula : 5 (in which R4 is hydrogen, ethoxy, mono(or di)lower 10 alkylamino, di(lower)alkylamino(lower)alkylamino, heterocyclicamino which may have suitable 15 substituent(s), or heterocyclic group which may have suitable substituent(s)); or a group of the formula : 20 25

or a salt thereof with a compound of the formula :



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to give a compound of the formula :

 $R^1$  N  $R_g^3$ 

wherein  $R^1$ ,  $R^2$  and  $R_g^3$  are each as defined above, or a salt thereof, or

(9) reacting a compound of the formula:

wherein  $R^1$  and  $R^2$  are each as defined above, and  $R^6$  and  $R^7$  are each lower alkyl, or a salt thereof with a compound of the formula :

25

NH  $H_2N-C-R^3$ 

wherein  $R^3$  is as defined above, or a salt thereof to give a compound of the formula :

5

25

$$R^{1}$$
 $N$ 
 $R^{3}$ 

wherein  $R^1$ ,  $R^2$  and  $R^3$  are each as defined above, or a salt thereof, or

(10) subjecting a compound of the formula:

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above and  $R_d^3$  is carboxy or protected carboxy, or a salt thereof to reduction reaction to give a compound of the formula :

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above,

or a salt thereof, or

(11) subjecting a compound of the formula:

wherein R<sup>1</sup>, R<sup>2</sup>, Y and Z are each as defined above, or a salt thereof to oxidation reaction to give a compound of the formula:

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above, or a salt thereof, or

(12) reacting a compound of the formula:

20

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above, or a salt thereof with a compound of the formula :

5 CH<sub>2</sub> COOF

or a salt thereof to give a compound of the formula :

10

R

Y
Z

N

CH=CH-COOH

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above, or a salt thereof, or

(13) reacting a compound of the formula:

25

R

Y
Z

NHCN

wherein  $\mathbb{R}^1$ ,  $\mathbb{R}^2$ , Y and Z are each as defined above, or a salt thereof with a compound of the formula :

35 HN R

wherein R<sup>8</sup> and R<sup>9</sup> are each hydrogen, lower alkyl, cyclo(lower)alkyl, ar(lower)alkyl, heterocyclic group which may have suitable substituent(s), or R<sup>8</sup> and R<sup>9</sup> are linked together with the attached nitrogen atom to form heterocyclic group which may have suitable substituent(s),

or a salt thereof to give a compound of the formula :

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wherein  $R^1$ ,  $R^2$ ,  $R^8$ ,  $R^9$ , Y and Z are each as defined above, or a salt thereof, or

(14) reacting a compound of the formula:

25 R<sup>1</sup> Y

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above, or a salt thereof with a compound of the formula :

NHCN

or a salt thereof to give a compound of the formula:

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above, or a salt thereof, or

(15) reacting a compound of the formula:

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above and  $x^2$  is a leaving group,

or a salt thereof with a compound of the formula :

$$\mathrm{HN} < \mathrm{R}^{10}$$

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wherein  $R^{10}$  is hydrogen or lower alkyl and  $R^{11}$  is lower alkyl or 1-amino-1-iminomethyl, or

R<sup>10</sup> and R<sup>11</sup> are linked together with the
 attached nitrogen atom to form
 heterocyclic group which may have
 suitable substituent(s),

or a salt thereof to give a compound of the formula :

wherein  $R^1$ ,  $R^2$ ,  $R^{10}$ ,  $R^{11}$ , Y and Z are each as defined above,

or a salt thereof, or

5

(16) reacting a compound of the formula:

25 R<sup>1</sup> CH<sub>3</sub>

wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above, or a salt thereof with a compound of the formula :

35 нсно

and a compound of the formula:

HN < R12

5

wherein  $\mathbf{R}^{12}$  and  $\mathbf{R}^{13}$  are each lower alkyl, or a salt thereof to give a compound of the formula :

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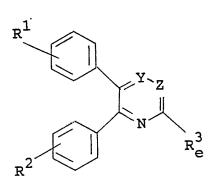
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wherein  $R^1$ ,  $R^2$ ,  $R^{12}$ ,  $R^{13}$ , Y and Z are each as defined above, or a salt thereof, or

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(17) subjecting a compound of the formula:

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wherein  $R^{1}$ ,  $R^{2}$ , Y and Z are each as defined above and  $R_{e}^{3}$  is carboxy(lower)alkenyl,

or a salt thereof to reduction reaction to give a compound of the formula:

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wherein  $R^1$ ,  $R^2$ , Y and Z are each as defined above and  $R_f^3$  is carboxy(lower)alkyl, or a salt thereof.

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- 10. A pharmaceutical composition which comprises, as an active ingredient, a compound of claim 1 or a pharmaceutically acceptable salt thereof in admixture with pharmaceutically acceptable carriers.
- 11. A use of a compound of claim 1 or a pharmaceutically acceptable salt thereof as antithrombotic, vasodilating or anti-inflammatory agent.
  - 12. A method for the prophylactic or therapeutic treatment of thrombosis, hypertension, cardiovascular or cerebrovascular diseases, or inflammation which comprises administering a compound of claim 1 or a pharmaceutically acceptable salt thereof to human or animals.
- 13. A process for preparing a pharmaceutical composition
  which comprises admixing a compound of claim 1 or a
  pharmaceutically acceptable salt thereof with a
  pharmaceutically acceptable carrier.

## INTERNATIONAL SEARCH REPORT

International Application No PCT/JP 91/01042

International Application No PCI/OP 91/01042									
I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) <sup>6</sup>									
According to international Patent Classification (IPC) or to both National Classification and IPC IPC5: C 07 D 253/065, 07, 403/04, 06, 12, 213/44, 54, 241/12, 20,									
IPC5:	220/26	253/065, 0/,	403/04, 06,	12, 213/44,	54, 241/12	, 20,			
II EIELI	239/26 DS SEARCH	42. 401/04.	06, 12, 413	704. 06. 12.	A 61 K 31/	395			
	JO OCANOI		Minimum Docum	nentation Searched 7					
Classifica	tion System			Classification Symbo	ois				
			· · · · · · · · · · · · · · · · · · ·	***************************************					
IPC5		C 07 D; A 6	1 K						
Documentation Searched other than Minimum Documentation									
				nts are included in Fig					
III. DOC	JMENTS CO	ONSIDERED TO BE R	ELEVANT <sup>9</sup>						
Category *	Citat	on of Document, <sup>11</sup> wit	th indication, where a	ppropriate, of the rele	vant passages <sup>12</sup>	Relevant to Claim No.13			
X	US, A,	3979516 (W.I	B. LACEFIELD	) 7 September	1976.	1-4,6,9-			
	SE	e the whole (	document	•	•	10,13			
Α						5,7,8			
X	IIS A	4021553 (W.E	RIACETEIN	ET AL \		1_4 0_			
^		May 1977,	J. LACEFIELD	CI AL.)		1-4,9- 10,13			
		e the whole o	document			10,15			
A						5-8			
v		4010011 (11.1			_				
Х		4318911 (W.E		) 9 March 198	32,	1-4,9-			
A	Se	e the whole o	ocument			10,13 5-8			
•						3-6			
			***						
:									
	_	es of cited document		"T" later docume	ent published after t	he international filing date ct with the application but			
con	_	ing the general state o e of particular relevan		invention	erstand the principle	or theory underlying the			
"E" earlier document but published on or after the international "X" document of particular relevance, the claimed invention									
"L" document which may throw doubts on priority claim(s) or involve an inventive step									
CITZ	tion or otne	r special reason (as si	pecified)	cannot be co	nsidered to involve	, the claimed invention an inventive step when the			
"O" doci									
"P" document published prior to the international filing date but "&" document member of the same patent family									
IV. CERTIFICATION									
Date of the Actual Completion of the International Search  Date of Mailing of this International Search Report									
3rd October 1991 0 5 NOV						10V 1991			
International Securities Authority									
internations	nternational Searching Authority Signature of Authorized Officer								
EUROPEAN PATENT OFFICE MISS T. TAZEL						TAZELAAR			
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Form PCT/ISA/210 (second sheet) (January 1985)

III. DOCL	MENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)	Relevant to Claim No
Category *	Citation of Document, with indication, where appropriate, of the relevant passages	Relevant to ordina its
Х	WO, A1, 8904308 (TERUMO KABUSHIKI KAISHA) 18 May 1989,	1-4,9- 10,13
4	see examples 14-15 and table 1	5-8
X	Chemical Abstracts, volume 104, no. 1, 6 January 1986, (Columbus, Ohio, US), Pitet G. et al.: "Synthesis and pharmacological profile of new diaryl-as-triazines", see page 18, abstract 228w, & Bull. Chim. Farm. 1985, 124(6), 271-	1-4,9- 10,13
A	278 	5-8
х	FR, A, 2173868 (MCNEIL LABORATORIES, INCORPORATED) 12 October 1973, see examples 41-45, pages 32-33 and claim	1-4,9- 10,13
A	14 	5-8
A	EP, A2, 0088593 (ELI LILLY AND COMPANY) 24 September 1983, see pages 1-2	1-10, 13

Form PCT/ISA/210 (extra sheet) (January 1985)

FURTHER INFORMATION CONTINUED FROM THE SECOND SHEET				
V.[X] OBSERVATIONS WHERE CERTAIN CLAIMS WERE FOUND UNSEARCHABLE 1				
This international search report has not been established in respect of certain claims under Article 17(2) (a) for the following reasons:  1. X Claim numbers 11-12 because they relate to subject matter not required to be searched by this Authority, namely:				
1.[X] Claim numbers <del>自由,自中</del> because they relate to subject matter not required to be searched by this Authority, namely:				
A method for treatment of the human or animal body by therapy,				
see rule 39.				
2.X Claim numbers 1-3, because they relate to parts of the international application that do not comply with the prescribed require-				
ments to such an extent that no meaningful international search can be carried out. pacifically:				
The scope of claims 1-3 is so broadly formulated that a very wide				
range of structures are included. These claims has thus not been				
fully searched.				
3. Claim numbers because they are dependent claims and are not drafted in accordance with the second and third sentences of				
3. Claim numbers, because they are dependent claims and are not drafted in accordance with the second and third sentences of PCT Rule 6.4(a).				
VI. OBSERVATIONS WHERE UNITY OF INVENTION IS LACKING 2				
This International Searching Authority found multiple inventions in this international application as follows:				
1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims				
of the International application.				
2. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims of the international application for which fees were paid, specifically claims:				
3. No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to				
the invention first mentioned in the claims; it is covered by claim numbers:				
4. As all searchable claims could be searched without effort justifying an additional fee, the International Searching Authority did not				
invite payment of any additional fee.  Remark on Protest				
The additional search fees were accompanied by applicant's protest.				
No protest accompanied the payment of additional search fees.				

## ANNEX TO THE INTERNATIONAL SEARCH REPORT ON INTERNATIONAL PATENT APPLICATION NO.PCT/JP 91/01042

SA

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This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on 30/08/91 The European Patent office is in no way liable for these particulars which are merely given for the purpose of information.

Patent document cited in search report		Publication date	Patent family member(s)		Publication date
US-A-	3979516	07/09/76	US-A- US-A- US-A- US-A-	3948894 3989831 4013654 4008232	06/04/76 02/11/76 22/03/77 15/02/77
 US-A-	4021553	03/05/77	NONE		
US-A-	4318911	09/03/82	NONE		
WO-A1-	8904308	18/05/89	EP-A- JP-A- JP-A- JP-A-	0397859 1128971 1128972 1135775	22/11/90 22/05/89 22/05/89 29/05/89
FR-A-	2173868	12/10/73	NONE		
EP-A2-	0088593	24/09/83	AU-B- AU-D- CA-A- GB-A-B- JP-A- US-A- US-A-	547581 1202983 1195327 2116179 58162582 4513135 4585861	24/10/85 08/09/83 15/10/85 21/09/83 27/09/83 23/04/85 29/04/86